

PPSC Physics Topic 3 Thermal Properties of Matter

Sr	Questions	Answers Choice
1	A fixed mass of an ideal gas absorbs 1000 J of heat and expands under a constant pressure of 20 kPa from a volume of $25 \times 10^{-3} \text{ m}^3$ to a volume 50×10^{-3} The change internal energy of the gas is.	A. 500 J B. 1000 J C. -1000 J D. Zero
2	Why an even Carnot engine Carnot give 100% efficiency.	A. We cannot find ideal sources B. We cannot eliminate friction C. We cannot reach absolute zero temperature D. We cannot remove heat
3	if temperature eon Celsius scale is 50 oC the temperature on Fahrenheit scale will be.	A. 102 ^o F B. 108 ^o F C. 112 ^o F D. 122 ^o F
4	The pressure of a gas is directly proportions to	A. Mean velocity of the molecules B. Mean square velocity of the molecules C. Root mean square velocity of the molecules D. Instantaneous velocity of the molecules
5	Which of the following should not change in an Isothermal process.	A. Volume B. Pressure C. Temperature D. All of these
6	A heat engine can develop efficiency equal to 100% if the temperature of the sink is	A. Less than that of source B. Equal to that of source C. 0 K D. 0 ^o C
7	A standard fixed point for calibrating a thermometer is.	A. Boiling point of water B. Melting point of ice C. Temperature of steam D. Triple point of water
8	If we place oure hand below a lighted lamp we feel warmer due to.	A. Conduction B. Convection C. Radiation D. None of these
9	Which law states that a change in the internal energy of a closed thermodynamic system is equal to the difference between the heat supplied to the system and teh amount of work by the system on the surrounding.	A. Zeroth law of thermodynamics B. First law of thermodynamics C. Second law of thermodynamics D. Third law of thermodynamics
10	A heat engine with 100% efficiency would have to	A. Do no work B. Be at a uniform temperature C. Use no heat D. Discharge at 0 ^o C
11	For which process is the relation $\Delta Q = \Delta V$ true.	A. Isothermal B. Adiabatic C. isobaric D. Isochoric
12	If T1 and T2 are source and sink temperature respectively Carnot efficiency is.	A. T_1+T_2/T_1 B. T_1-T_2/T_1 C. T_1+T_2/T_2 D. T_1-T_2/T_2
13	Law of increase of entropy is a result of	A. First law of thermodynamics B. Second law of thermodynamics C. Third law of thermodynamics D. Zeroth law of thermodynamics
14	When the temperature of source and sink of a heat engine become equal the entropy change will be.	A. Zero B. Maximum C. Minimum D. Negative

15	On which parameter, the heat capacity of a material depends upon.	A. Density of the material B. Specific heat of the material C. Temperature of the material D. Structure of the material
16	Which thermometer is called spirit thermometer	A. Alcohol thermometer B. Mercury in glass thermometer C. Gas thermometer D. Radiation thermometer
17	The actual gas can behave like an ideal gas at	A. Low density and high pressure B. High density and high pressure C. Low density and low pressure D. High density and low pressure
18	On which of the following the kinetic theory of gases is not applicable.	A. Water vapour B. Smoke particles C. Bound particles D. Free electrons
19	Which kind of motion is exhibited by molecules of monoatomic gas.	A. Rotatory B. Vibratory C. Translatory D. Random
20	The gas thermometer is taken as the primary standard because.	A. Thermometers are easily reproducible B. Readings can be accurately taken C. No correction are necessary D. It produces the thermodynamic scale