

PPSC Chemistry Part III Inorganic Chemistry Online Test

Sr	Questions	Answers Choice
1	iodine is used as	A. Tincture of iodine B. Iodex and antiseptic C. Treatment of goiter D. All above
2	Iodine is a grey black solid and its vapours are in color	A. Grey B. Black C. Yellow D. Violet
3	Bromine is used as	A. Fungicides B. Herbicides C. Germicides D. Insecticides
4	Bromine is soluble in	A. Alcohol B. Water C. Chloroform D. All above
5	The vapours attacks the eyes and mucous membrane of nose and throat	A. F B. Cl C. I D. Br
6	The bromine produced on commercial scale may contain impurities of.	A. Water B. Chloride C. Iodine D. All above
7	Chlorine is used in	A. Sterilization of water B. Extraction of gold C. Bleaching of cotton D. All above
8	Used for sterilization of drinking water	A. F B. Br C. Cl D. I
9	Greenish yellow gas with pungent irritating odour	A. Chlorine B. Fluorine C. Iodine D. Bromine
10	On industrial scale chlorine is prepared by	A. Dennis method B. Deacon's process C. Plantner's process D. Aludels process
11	The electrolytic method super passes all other methods due to.	A. Furity B. Cheapness C. Easy available D. All above
12	Separation of isotopes of uranium is carried out by	A. CaF_2 B. SF_6 C. HF D. All above
13	Fluorine finds considerable use of DDFT which is used as.	A. herbicide B. Fungicide C. Insecticide D. Nematocides
14	Fluorine is.	A. Powerful oxidizing agent B. Most reactive element C. Used as refrigerants D. All of above
15	Fluorine form Fluorides reacting with	A. Metals B. Non metals C. Metalloids D. Any of above

16	In Dannis's method the end of the copper caps into which graphite electrode are fixed with cement.	A. Portiant B. Bakelite C. Asbestos D. All of above
17	Among all halogens no oxyacid of the following is known	A. F B. Cl C. Br D. I
18	Flourine differs from the other members of its own group due to.	A. Its small size and low bond energy B. Its higher electornegativity C. None-availability of d-orbitals in its valence shell D. All the above
19	All the halogen form oxyacide, except	A. Florine B. Chlorine C. Bromine D. Iodine
20	In their ionic compounds halogens exhinit the oxidation states of.	A. -1 B. -2 C. -3 D. -4