

PPSC Chemistry Part III Inorganic Chemistry Online Test

Sr	Questions	Answers Choice
1	Nitric acid is used in manufacturing of.	A. Explosive B. H_2SO_4 C. Fertilizer D. All above
2	H_2SO_4 is manufactured by	A. The lead chamber process B. The contact process C. Both A and B D. The Ostwald's process
3	The noble gases are used due to having property	A. Chemical inertness B. Low boiling point C. Any of a or b D. Both a and b
4	Elements in the same vertical group of the periodical have same	A. Number of electron B. Atomic number C. Number of valence electrons D. Electronic configuration
5	Molecule of oxygen is	A. Diamagnetic B. Paramagnetic C. Both A and B D. None of above
6	Which among the following is a false statement.	A. SiO_2 has a structure similar to that of CO_2 B. Natural Si exists only in the combined state C. Si can be prepared by reducing SiO_2 with Mg D. Si does not exist in graphite like structure, but exists only in diamond like structure.
7	CFSE for d^7 ion is.	A. 0.8 B. -0.8 C. -1.8 D. 1.8
8	In urea the amount of nitrogen is	A. 82.0% B. 46.0% C. 33.0% D. 21.0%
9	Any substance which has solidified from the liquid state with crystallization is known as	A. Steel B. Fibre C. Glass D. Asbestos
10	Since the acid gives both acidic and normal salts so the acid is.	A. di acid B. di basic C. double salt D. Any of above
11	Stainless steel contains	A. Fe+Cr+Ni B. Fe+Ni+Cu C. Fe + Cr+ Cu D. Cu + C + Ni
12	Elements of group 14 have the electronic configuration of their outer shell as	A. $\text{ns}^2 \text{np}^3$ B. $\text{ns}^2 \text{np}^2$ C. $\text{ns}^2 \text{np}^6$ D. ns^2
13	The magnitude of electron affinity depends on.	A. Atomic size B. Nuclear charge C. Electronic configuration D. All of the above
14	Which one of the following has the biggest electron affinity.	A. F_2 B. Cl_2 C. Br_2 D. I_2

15 The penultimate shell of carbon contains electrons.
A. s2
B. s2p6
C. s2p6d10
D. s2p6d8

16 Which of the following has the highest melting point.
A. NaCl
B. LiCl
C. KCl
D. RbCl

17 Nitric acid has the property
A.

Nitrating

B. Reducing
C. Redoxing
D. None of above

18 Which of the ionic possesses highest bond energy.
A. C-C
B. Si-Si
C. Ge-Ge
D. Sn-Sn

19 Which of the following compounds has highest boiling point.
A. HI
B. HF
C. HBr
D. HCl

20 Sodium silicate is used
A. In fire proofing of wood and textiles
B. As a preservative of eggs
C. As a furniture polish
D. All above
