

PPSC Chemistry Part II Organic Chemistry Online Test

Sr	Questions	Answers Choice
1	Which of the following class of compounds follow the criteria of aromaticity.	A. The compounds must have high degree of unsaturation B. they must have the property to undergo addition reactions C. They must have the property to undergo substitution reactions D. They must have the ability to sustain an induced current in NMR
2	IUPAC name of HCONH_2 is.	A. Methanamide B. Methanoylamine C. Ammonoethanal D. Formanide
3	α - pinene hydrochloride on warming rearranges to form bornyl chloride. The rearrangement is known as.	A. Pinacol pinacolone B. Hofmann C. Wager Meierwein D. Wolff
4	Toluene is o/p -orienting with respect to an electrophilic substitution reaction due to.	A. +I effect of the methyl group. B. +I as well as +H effect of the methyl group C. Hyper conjugation between the methyl group and phenyl ring. D. + R effect of the methyl group
5	Which of the following statement is false about resonance.	A. It increases the stability of a molecule B. It leads to similar type of bonds C. It increases the reactivity of the molecule D. It decreases the reactivity of the molecule.
6	Compounds HCN and HNC are.	A. Tautomers B. Metamers C. Functional isomers D. Conformers
7	A terpenoid which has an alcoholic group in the molecule is	A. Citral B. Camphor C. Menthol D. Carvone
8	Arrangement of peptide chains of protein in such a way to form helix structure is referred to as.	A. Primary structure B. Secondary structure C. Tertiary structure D. Quaternary structure
9	Monomers are Teflon is	A. Monochloroethene B. 1,2- Difluoroethene C. 1,1,2- Trifluoroethene D. Tetrafluoroethene
10	Which of the following statements is not correct, with respect to resonance.	A. The position of atomic nuclei must be same B. The limiting structures must have same number of paired and unpaired electrons. C. The energy of the various limiting structures must contribute equally D. All above
11	Sugar and common salt in a mixture can be separated through the process of.	A. Sublimation B. Distillation C. Ion exchange D. Crystallization from solution in ethanol
12	Nitrobenzene can be prepared from benzene by using a mixture of conc. HNO_3 and conc. H_2SO_4 in the nitrating mixture. HNO_3 acts as a.	A. Base B. Acid C. Oxidizing agent D. Catalyst

13	α -terpineol is obtained on hydration of which of the following with dilute H_2SO_4 .	B. Myrcene C. Linalool D. Limonene
14	Which of the following is a triphenylmethane dye.	A. Auramine G B. Crystal violet C. Fluorescein D. Fast green O
15	Hydrolysis of protein gives	A. α -amino acid only B. β -amino acids only C. gamma amino acid only D. A mixture of all of these
16	Which of the following molecules can exhibit geometrical isomerism.	A. $\text{CH}_3\text{CH}=\text{CH}_2$ B. $\text{CH}_3\text{CH}=\text{CHCH}_3$ C. $(\text{CH}_3)_2\text{C}=\text{CH}_2$ D. $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)_2$
17	Lactic acid is a molecule which shows	A. Epimerism B. Tautomerism C. Optical isomerism D. Metamerism
18	Among the following statements in the nitration of aromatic compounds, the false one is.	A. The rate of nitration of benzene is almost the same as that of hexadeutero benzene B. The rate of nitration of toluene is greater than that of benzene C. The rate of nitration of benzene is greater than that of hexadeutero benzene. D. Nitration is an electrophile substitution reaction.
19	Different arrangement of groups in space which can be converted into one another by rotation around a single bond are called.	A. Conformations B. Metameres C. Enantiomers D. All of the above
20	The greater stability of benzyl carbonium ion as compared to t-butyl carbonium ion is due to.	A. Inductive effect B. Resonance effect C. Electrometric effect D. All above