

PPSC Chemistry Full Book Test

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| Sr | Questions | Answers Choice |
| 1 | Which is the correct order of wave number of the following redistions. | A. X-rays > uv > Infrared > visible> radio waves B. X-rays > uv > visible > Infrared > radio waves C. X-rays > radio waves > uv > visible > Infrared D. X-rays > Infrared > uv> visible > radio waves |
| 2 | In 1952 who popularized the use of CFT for inorganic chemist | A. Bethe B. Orge C. Van Vleck D. Werner |
| 3 | Among the solvents given below, with dielectric constant (E) given in parentheses which has highest solubility of KCI? | A. Benzene (E=O) B. Carbon disulphide (E = O) C. Methanol (E =32) D. Acetone (E = 2) |
| 4 | Which of the following analytical technique is used for the separation of an interfering substance or analyte from the mixture. | A. Precipitation B. Distillation C. Electrode position D. All above these |
| 5 | Which of the following is renewable energy source. | A. Moon B. Wind C. Sun D. Ocean |
| 6 | Which of the following ions is smallest in size. | A. F- B. Cl- C. I- D. Br- |
| 7 | Which of the following statements is correct. | A. A sigma bond is weaker than a pi pond B. There are four coordinate bonds in the Lewis structure of NH4+ ion. C. The 1 covalent bond is directional in nature D. A single bond between the two atoms cannot be re bond. |
| 8 | Which of the following molecules belongs to C_{av} point group. | A. H2O B. H2S C. NH3 D. BF3 |
| 9 | The diameter of fly ash particles is micro meter | A. 5-10 B. 10-20 C. 20-30 D. 100 |
| 10 | The principle former of almost all glasses is | A. (SIO2)n B. (SiO3)n C. (SiO2) D. None of these |
| 11 | Which parameter of a chemical reaction will change with the use of a catalyst. | A. Delta F, change in free energy B. Delta S, change in entropy C. Delta E, change in internal energy D. K, the rate constant |
| 12 | A minor constant is one whose amount in the sample is | A. 0.1 to 1 % B. 0.01 to 1% C. 1 to 10% D. None of the above |
| 13 | The rate constant of a reaction has same units as the rate of the reaction The reaction is of. | A. Second order B. First order C. Three order D. Zero order |
| 14 | The bond length is measured by | A. X-ray diffraction B. Neutron diffraction |

| | , | C. Microwave spectroscopy D. All of above |
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| 15 | The sugar present in DNA is | A. D- Ribose B. D-Glucose C. 2- Doxy D-Ribose D. 3-Deoxy D-ribose |
| 16 | For each value of I. the number of m velocity are. | A. n ² B. 2l C. (2l+1) D. (n+1) |
| 17 | Trimethylamine is a weaker base than dimethylamine is explained by | A. Steric effct B. Resonance effect C. Inductive effect D. All above |
| 18 | Which of the following elements would have the lowest first ionization energy | A. Mg B. Rb C. Li D. Ca |
| 19 | Which of the following statement is not correct with respect to electrometric effect. | A. It is permanent effect B. It is brought into play instantaneously at the demand of attacking reagent C. It proceeds a polar addition reaction D. The original electronic condition is restored after the removal of tacking regent. |
| 20 | Which of the following radical is a member of VI group. | A. Mg ²⁺ B. Na+ C. K+ D. NH ₄ ⁺ E. All above |