

Physics ICS Part 2 Chapter 20 Online MCQ's Test

Sr	Questions	Answers Choice
1	First spectral series of hydrogen atom was discovered by	A. Lyman B. Rydberg C. Balmer D. Paschen
2	The velocity of electron moving in 1st orbit of hydrogen atom is:	A. $2.09 \times 10^6 \text{ ms}^{-1}$ B. $2.18 \times 10^6 \text{ ms}^{-1}$ C. $2.19 \times 10^6 \text{ ms}^{-1}$ D. $3.18 \times 10^6 \text{ ms}^{-1}$
3	The longest wavelength of Paschen series is.	A. 656 nm B. 1094 nm C. 1875 nm D. 2000 nm
4	Charge on positron is:	A. Negative B. Positive C. Neutral D. None of these
5	Which series lies in the ultraviolet region.	A. Balmer series B. Bracket series C. Paschen series D. Lyman series
6	The first laser was built by	A. ArthursSchawalow B. T.H.Maiman C. Peter Sorokin D. C.H.Townes
7	Frequency of x-rays depends upon.	A. Number of electrons striking target B. Accelerating potential C. Nature of the target D. Both B and C
8	The idea of laser device was first introduced by C.H. Townes and Authers Schowlan is	A. 1972 B. 1965 C. 1958 D. 1913
9	The process by which lesser beam can be used to generate 3-dimensional images of objects is called	A. Holography B. Geo graphy C. Tomography D. Radio graphy
10	In according with Bohr's theory the K.E of the electron is equal to:	A. $ke^2/2r$ B. $Ze^2/2r$ C. $Ze^2/2r/r^2$ D. $Ze^2/2r^2$
11	The first spectral lines were discovered in 1885, were	A. Paschen series B. Balmer series C. Pfund series D. Bracket series
12	Balmer Empirical formula explains the electromagnetic radiation of any excited atom in terms of their.	A. Energy B. Mass C. Wave length D. Momentum
13	In an electronic transition atom cannot emit.	A. Infrared radiations B. Visible radiations C. Ultraviolet radiations D. Gama radiations
14	For an atom of hydrogen atom the radius of the first orbit is given by:	A. H/m^2 B. $me/4h^2$ C. $h^2/4\pi^2kme$

15 The radius of 10th orbit in hydrogen atom is.
A. 0.053 nm
B. 0.53 nm
C. 5.3 nm
D. 53 nm

16 The temperature of core of nuclear reactor is:
A. 1100°C
B. 1200°C
C. 1300°C
D. 1400°C

17 In Helium Neon laser, the discharge tube is filled with
A. 80% He, 20% Neon
B. 85% He, 15% Neon
C. 83% He, 17% Neon
D. 90% He, 10% Neon

18 Electron volt is unit of:
A. Chemical energy
B. Potential energy
C. Nuclear energy
D. heat energy

19 For Holography we use
A. X ray
B. Laser
C. gama rays
D. Beta rays

20 The line radiations emitted from by hydrogen filled discharge tube can be analyzed into.
A. Band spectrum
B. Line spectrum
C. Continuous spectrum
D. Absorption spectrum