

## Physics ICS Part 1 Chapter 6 Online Test

Sr	Questions	Answers Choice
1	The change in internal energy is defined as.	A. <p>Q - W</p> B. <p>Q - T</p> C. <p>Q + P</p> D. <p>Q - P</p>
2	When the system is expanded by adding heat energy, then the work done will be	A. Positive and on the system B. Negative and on the system C. Positive and by the system D. Negative and by the system
3	Change in entropy is maximum when temperature of source is that.....of sink	A. Greater than B. Less than C. Equal to D. Zero
4	When hot and cold water are mixed the entropy	A. Decrease B. Increase C. Remains constant D. Zero
5	Work done by the system is taken as	A. Positive B. Negative C. Undefined D. None of these
6	What can be calculated from the curve under PV graph.	A. Heat B. Work done C. Temperatures D. Internal energy
7	In Carnot engine, each process is.	A. Reversible B. Preferable Reversible C. Irreversible D. Perfectly irreversible
8	When two objects are made in thermal contact having same temperature then they are at.	A. Thermal Equilibrium B. Chemical equilibrium C. Mechanical Equilibrium D. Physical Equilibrium
9	In all natural processes where heat flow from one system to another there is always a net increase in	A. Pressure B. Entropy C. Work D. Volume
10	A system does 600 J of work and at the same time has its internal energy increased by 320 J. How much heat has been supplied.	A. 920 J B. 280 J C. 600 J D. 200 J
11	Internal energy is similar to the	A. Vibrational K.E. B. Gravitational P.E. C. K.E. D. All of these
12	The sum of all forms of molecular energies of substance is termed as	A. Kinetic energy B. Potential energy C. Internal energy D. Heat energy
13	'R' is called	A. Universal constant B. Universal per molecule constant C. Universal gas constant D. All of the above
14	The process which is carried out at constant temperature is called.	A. Adiabatic process B. Isothermal process C. Isochoric process D. Isobaric process

15 Efficiency of a carnot engine is.

A. <p>Infinite</p>  
B. <p>Ten</p>  
C. <p>Greater than 1</p>  
D. <p>Less Than</p>

---

16 The efficiency of carnot eninge is always.

A. <p>Greater than real engine</p>  
B. <p>Less than real engine</p>  
C. <p>Equal to the real engine</p>  
D. <p>Both a and b</p>

---

17 Collision between gas molecules are perfectly

C

A. <p>&nbs; Elastic</p> <p>E</p>  
B. <p>&nbs;Inelastic</p>  
<p>&nbs;</p>  
C. <p>&nbs; Neither elastic nor  
inelastic</p> <p>&nbs;</p>  
D. <p>&nbs; All of these</p>  
<p>&nbs;</p>

---

18 According to kinetic theory of gases, the size of the molecule is.

A. <p>&nbs; Much smaller than the  
separation between molecule</p>  
<p>M</p>  
B. <p>&nbs; Much larger than the  
separation btween molecules</p>  
<p>&nbs;</p>  
C. <p>&nbs; Both a and b</p>  
<p>&nbs;</p>  
D. <p>&nbs; Much larger ahnt the  
separation between atom</p>  
<p>&nbs;</p>

---

19 First law of thermodynamics is based upon law of conservatio of.

A. <p>Mass</p>  
B. <p>Momentum</p>  
C. <p>Energy</p>  
D. <p>Charge</p>

---

20 What one is not an isothermal change.

A. <p>Melting of solid</p>  
B. <p>Bioling of liquid</p>  
C. <p>Bursting of bicycle tyre</p>  
D. <p>Slow expansion of gas</p>