

Everyday Science Online Test

Sr	Questions	Answers Choice
1	Which elements are called bridge elements?	A. Element of the third period B. Element of he second period C. Element of the first group D. Element of the second group C+>
2	Which of the these is not a noble gas	A. Argon B. Hydrogen C. Krypton D. Neon D. Neon D. Neon
3	Who disproved the 'plum pudding' model of the atom?	A. Dalton b. Cavendish C. Ernest Rutherford D. Bohr
4	What is the only element that normally has no neutrons?	A. Oxygen B. Hydrogen C. Krypton D. Neon C. Krypton
5	Who disproved the Rutherford model of the atom?	A. Dalton b. Cavendish C. Langumir D. Bohr b.
6	What is an empirical formula?	A. the mean mass of the atoms of an element B. The simplest ratio of atoms of each element in a compound C. the sum of protons and neutrons in the nucleus D. The actual numbers of atoms of each element in a compound br>
7	German silver is an alloy of	A. Copper: 55 nickel: and zinc: 20 B. copper: 25 nickel: 45 and zinc: 30 C. copper: 25 nickel: 55 and zinc: 20 D. copper: 15, nickel: 45 and zinc: 40 Dr>
8	What is the percentage of silver in German Silver (also known as nickel Silver)	A. 100 B. 0 C. 50 D. 25
9	Which of the these metals has the highest density?	A. Platinum B. Lithium C. Tungsten D. Iridium b>
10	Which one of the following metals does not react with water produce hydrogen:	A. Sodium B. Potassium C. Lithium D. Cadmium D. Cadmium D. Cadmium D. Cadmium D. Cadmium D. Cadmium D. Cadmium