

Chemistry Fsc Part 2 Chapter 5 Online Test

Sr	Questions	Answers Choice
1	Which halogen will react spontaneously with Au(s) to produce Au ³⁺	A. Br ₂ B. F ₂ C. I ₂ D. Cl ₂
2	Which acid can not be stored in glass bottles.	A. HCI B. HF C. H2SO4 D. HNO3
3	Hydrogen bond is the strongest between the molecules of.	A. HF B. HCI C. HBr D. HI
4	Which halogen will react spontaneously with Au to produce Au3+	A. Br2 B. F2 C. I2 D. CI2
5	In fluorescent tube, the gas filled is.	A. He B. Ne C. Ar D. Xe
6	Which one of the following noble gases is least polarizable	A. He B. Ne C. Ar D. Kr
7	Colour of which halogen is not correctly related.	A. F2 colourless gas B. Cl2 greenish yellow gas C. Br2 Reddish brown liquid D. l2 grayish Black solid
8	Bleaching powder contains available chlorine approximately	A. 100% B. 70-80% C. 35-40% D. 10-20%
9	Hydrogen bond is the strongest between the molecules of	A. HF B. HCI C. HBr D. HI
10	Bleaching powder may be produced by passing chlorine over.	A. Calcium carbonate B. Hydrated calcium sulphate C. Anhydrous calcium sulphate D. Calcium hydroxide
11	Which is the strongest acid	A. HCIO B. HCIO ₂ C. HCIO ₃ D. HCIO ₄
12	The anhydride of HClO4 is	A. CIO3 B. CIO2 C. CI2O5 D. CI2O7
13	XeF6 n hydrolysis produces.	A. XeOF2 B. XeOF3 C. XeOF4 D. XeF2
14	The chemical formula of Sodium Bromite is.	A. NaBrO B. NaBrO2 C. NaBrO3 D. NaBrO4
15	Which substance is used in photography	A. AgCl B. AgBr C. AgI D. Ag3PO4

16	Bleaching powder may be produced by passing chlorine over	A. calcium carbonate B. hydrated calcium sulphate C. calcium hydroxide D. magnesium hydroxide
17	is use as a cooling medium for nuclear reactors	A. Ne B. He C. Ar D. Kr
18	Which one of the following is not use of chlorine.	A. Formation PVCB. Formation of mustard gasC. Disinfectant and bleaching agenD. Formation of sodium chloride
19	lodine deficiency in diet is known to cause.	A. Beri Beri B. Goiter C. Rickets D. Night blindness
20	Which of the following hydrogen halide is the weakest acid in solution.	A. HF B. HBr C. HI D. HCI