

Chemistry Fsc Part 2 Chapter 4 Online Test

Sr	Questions	Answers Choice
1	Which catalyst is used in contact process.	A. Fe ₂ O ₃ B. V ₂ O ₅ C. SO ₃ D. Ag ₂ O
2	The brown gas formed when metal reduces HNO ₃	A. N ₂ O ₅ B. N ₂ O ₃ C. NO ₂ D. NO
3	The most electronegative element of group V-A is	A. N B. P C. Sb D. Bi
4	In group VA elements the most electronegative elements is.	A. Sb B. N C. P D. As
5	Lowest oxidation state of nitrogen is present in.	A. NH ₃ B. NO ₂ C. NO D. HNO ₃
6	NO ₂ can be obtained by heating.	A. NaNO ₃ B. KNO ₃ C. Pb(NO ₃) ₂ D. NH ₃ NO ₃
7	Which one is metaphosphoric acid	A. HPO ₃ B. H ₃ PO ₃ C. H ₃ PO ₄ D. H ₄ P ₂ O ₇
8	In pyrite burner, the gas produced is.	A. SO ₃ B. SO ₂ C. CO ₂ D. NO
9	Which of the following gives acidic oxide	A. N B. As C. Sb D. Bi
10	When concentrated H ₂ SO ₄ and solid sodium chloride react together at room temperature the product are.	A. Two salts only B. A salt and a base C. A salt and an acid D. A salt and water
11	Role of H ₂ S in the given chemical reaction is H ₂ S + I ₂ ----- 2HI+S	A. Oxidising agent B. Reducing agent C. Dehydrating agent D. As an acid
12	In which substance phosphorus is not present.	A. Yolk of egg B. Bones C. Apatite D. Galena
13	What is %age of calcium phosphate in bone ash	A. 20 B. 40 C. 80 D. 60
14	Which one of the following group of Periodic table called chalcogen family.	A. Group III A B. Group VA C. Group VI A D. Group VII A
15	Arsenic impurities in contact process are removed.	A. By prolong heating the gases B. By treatment with Fe(OH) ₃ C. In scrubbing tower D. In absorption tower

16	Which of the following is a reddish brown gas	A. N_2O_3 B. NO_2 C. N_2O_3 D. N_2O_5
17	Basicity of ortho phosphoric acid is.	A. 1 B. 2 C. 3 D. 4
18	Out of all the elements of Group V-A the highest ionization energy is possessed by	A. N B. P C. Sb D. Bi
19	Which of the following acids possess oxidizing and reducing properties.	A. HCl B. HNO_2 C. HNO_3 D. H_2SO_4
20	Out of all the elements of group VA, the highest ionization energy is possessed by	A. N B. As C. Sb D. Bi