

Chemistry Fsc Part 2 Chapter 3 Online Test

Sr	Questions	Answers Choice
1	Litharge is chemically	A. PbO B. PbO ₂ C. Pb ₃ O ₄ D. Pb(CH ₃ COO) ₂
2	The aqueous solution of Borax	A. Acidic B. alkaline C. Amphoteric D. manual
3	Tincal is a mineral of	A. Al B. C C. Si D. B
4	The chief of aluminum is	A. Na ₃ AlF ₆ B. Al ₂ O ₃ ·2H ₂ O C. Al ₂ O ₃ D. Al ₂ O ₃ ·H ₂ O
5	Kaolin is a mineral of	A. Carbon B. Magnesium C. Silicon D. Aluminium
6	Aluminum reacts with nitrogen to form	A. AlN B. Al ₂ N C. Al ₂ N ₃ D. Al ₄ N ₆
7	Aluminium does not corrode because.	A. It is a semi metal B. It is a silver shining metal C. It does not react with O ₂ D. It forms a protective layer of Al ₂ O ₃
8	Aluminium oxide is	A. Acidic oxide B. Basic oxide C. Amphoteric oxide D. None of these
9	Which one of the followings is not ore of aluminium	A. Corundum B. Bauxite C. Colemanite D. Kaolin
10	The composition of white lead is	A. PbCO ₃ ·Pb(OH) ₂ B. PbCO ₃ ·2Pb(OH) ₂ C. 2PbCO ₃ ·Pb(OH) ₂ D. PbCO ₃
11	Which element among the following belongs to Group IV A of the Periodic table.	A. Barium B. Iodine C. Lead D. Oxygen
12	Third abundant element on earth crust is.	A. Boron B. Oxygen C. Aluminium D. Silicon
13	Aluminum oxide is	A. acidic oxide B. basic oxide C. amphoteric oxide D. none of these
14	Which is not used of Borax	A. Softening of water B. As medicine for washing eyes C. As flux in metallurgical operations D. To make quartz
15	The shape of SiO ₂ IS	A. Tetrahedral B. Trigonal bipyramid C. Linear D. Cubic

16	Boric acid is used as eye wash due to its	A. Weakly acidic property B. Antiseptic nature C. Antibiotic nature D. Weakly basic properties
17	In which substance silicon is not present.	A. Talc B. Asbestos C. Dolomite D. Zircon
18	Which of the following elements is not present abundantly in earth's crust.	A. Silicon B. Aluminium C. Sodium D. Oxygen
19	Lead monoxide is	A. Amphoteric B. Neutral C. Acidic D. Basic
20	Chemical composition of colemanite is	A. $\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 5\text{H}_2\text{O}$ B. $\text{CaB}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$ C. $\text{Na}_2\text{B}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$ D. $\text{CaNaB}_5\text{O}_9 \cdot 8\text{H}_2\text{O}$