

Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	Which of the following sulphates is not soluble in water.	A. Sodium sulphate B. Potassium sulphate C. Zinc sulphate D. Barium Sulphate
2	Which elements are deposited at the cathode during the electrolysis of brine in diaphragm cell	A. H ₂ B. Ba C. Ra D. Rn
3	Which metal oxide is insoluble in water	A. MgO B. CaO C. SrO D. BaO
4	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
5	Chile saltpeter has the chemical formal	A. NaNO ₃ B. KNO ₂ C. Na ₂ B ₄ O ₇ D. Na ₂ CO ₃ .H ₂ O
6	Which one of the following elements has the same oxidation number in all its known compounds.	A. Beryllium B. Nitrogen C. Sulphur D. Chorine
7	Down's cell is used to prepare.	A. Sodium carbonate B. Sodium bicarbonate C. sodium metal D. Sodium hydroxide
8	Which one of the following has lowest first ionization energy.	A. Li B. Na C. Rb D. Cs
9	Which oxide is amphoteric in nature.	A. BeO B. MgO C. CaO D. BaO
10	In Down's cell anode is made up of.	A. Iron B. Copper C. Graphite D. Platinum
11	Sodium is collected over molten sodium chloride in Down's cell, because.	A. It does not react with NaCl B. It is lighter in weight and floats on molten NaCl C. Its reactivity is greater than NaCl D. None of these
12	Ca Mg3 (SiO3)2 is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos
13	Which one of the following is not an alkali metal	A. Francium B. Caesuym C. Rubidium D. Radium
14	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
15	Which one of the following does not belong to alkaline earth metals.	A. Be B. Ra C. Ba

		D. Rn
_		A. Magnesite B. Epsom salt
16	In which ore carbon is present	C. Barite
		D. Sylvite
		A. O2
7	Which gas is evolved when CO2 is passed through KO2.	B. CO
•		C. O3
		D. None of these
	Which one is not use of lime	A. Extraction of metals
8		B. Ceramic industry
0		C. Preparation of bleaching powder
		D. Preparation of gypsum
	Which element reacts with alkali and H2 gas is produced.	A. Be
9		B. Mg
9		C. Ca
		D. Sr
	When water is added to plaster of paris, it changes to a hard mass. Its volume also expands upto	A. 1%
20		B. 2%
.0		C. 3%
		D. 5%