

Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be
2	The element cesium bears resemblance with	A. Ca B. Cr C. Both of the above D. None of the above
3	Alkali metal whose carbonate is relatively less stable and decomposes giving its oxides is.	A. Li B. Ba C. K D. Rb
4	Which of the following sulphates is not soluble in water.	A. Sodium sulphate B. Potassium sulphate C. Zinc sulphate D. Barium Sulphate
5	The oxides of beryllium are	A. Basicity B. Rancidity C. Amphoteric D. none of all
6	Which hydroxide decomposes on heating.	A. LiOH B. NaOH C. KOH D. CaOH
7	Which oxide is amphoteric in nature.	A. BeO B. MgO C. CaO D. BaO
8	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
9	In Down's cell anode is made up of.	A. Iron B. Copper C. Graphite D. Platinum
10	Which of the following gas will turn lime water milky	A. Cl ₂ B. NO ₂ C. CO D. CO ₂
11	Dolomite is a carbonate of	A. Be B. Mg C. Na D. Ba
12	Which ion will have the maximum value of heat of hydration.	A. Na ⁺ B. Ca ⁺ C. Ba ²⁺ D. Mg ²⁺
13	The most metallic elements from the following is	A. Arsenic B. Oxygen C. Antimony D. Bismuth
14	Which of the following compounds is used to make spray, which has fungicidal action.	A. BaSO ₄ B. Na ₂ CO ₃ C. CaO D. CaSO ₄ .2H ₂ O
15	When water is added to plaster of paris, it changes to a hard mass. Its volume also expands upto	A. 1% B. 2% C. 3% D. 5%

16 Which one of the following does not belong to alkaline earth metals.

A. Be
B. Ra
C. Ba
D. Rn

17 The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.

A. Gain water and remains solid
B. Gain water and becomes liquid
C. Lose mass and remain solid
D. Remain unchanged on heating.

18 Which one of the following does not belong to alkaline earth metals

A. Be
B. Ra
C. Ba
D. Rn

19 The Milk of magnesia is used for the treatment of

A. Basicity
B. Rancidity
C. Acidity
D. Jaundance

20 Sodium is collected over molten sodium chloride in Down's cell, because.

A. It does not react with NaCl
B. It is lighter in weight and floats on molten NaCl
C. Its reactivity is greater than NaCl
D. None of these
