

Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	The oxide of beryllium is	A. Acidic B. Basic C. Amphoteric D. None of these
2	Down's cell is used to prepare.	A. Sodium carbonate B. Sodium bicarbonate C. sodium metal D. Sodium hydroxide
3	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
4	Sulphur is essential constituent for plants. Which is not role of sulphur	A. Chlorophyll development B. Development of root system C. Constituents of some proteins D. Increase transpiration force
5	The word alkali is derived from which language	A. Arabic B. Greek C. French D. German
6	Which element reacts with carbon to form metal carbide.	A. Li B. Na C. K D. All of these
7	Which elements are deposited at the cathode during the electrolysis of brine in diaphragm cell	A. H ₂ B. Ba C. Ra D. Rn
8	Point out the element which forms super oxide	A. Li B. Na C. K D. C
9	When water is added to plaster of paris, it changes to a hard mass. Its volume also expands upto	A. 1% B. 2% C. 3% D. 5%
10	The Milk of magnesia is used for the treatment of	A. Basicity B. Rancidity C. Acidity D. Jaundance
11	Which is insoluble in water.	A. BeSO4 B. MgSO4 C. CaSO4 D. BaSO4
12	Which one does not belong to the alkaline earth metals	A. Be B. Ba C. Ra D. Rn
13	Which hydroxide decomposes on heating.	A. LIOH B. NaOH C. KOH D. CaOH
14	Sodium is collected over molten sodium chloride in Down's cell, because.	A. It does not react with NaCl B. It is lighter in weight and floats on molten NaCl C. Its reactivity is greater than NaCl D. None of these
15	does not belong to Alkaline-Earth metals	A. Be B. Ra C. Ba

		D. Kn
16	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	A. Gain water and remains solid B. Gain water and becomes liquid C. Lose mass and remain solid D. Remain unchanged on heating.
17	Which of the following compounds is used to make spray, which has fungicidal action.	A. BaSO4 B. Na2CO3 C. CaO D. CaSO4.2H2O
18	Which on eof the following alkaline earth metal form peroxide when heated with oxygen at $600\ ^{\mathrm{o}}\mathrm{C}$	A. Magnesium B. Calcium C. Strontium D. Barium
19	Chile saltpeter has the chemical formula	A. NaNO3 B. KNO3 C. Na2B4O7 D. Na2CO3.H2O
20	Setting of Plaster of Paris is accompanied with	A. Hydrogen bonding B. Hydration C. Dehydration D. None of these