

## Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
2	Which the correct statement	A. Cl <sup>-</sup> is smaller than Cl atom B. Cl <sup>(ion)</sup> and Cl (atom) are equal in size C. Na <sup>+</sup> is smaller than Na atom D. Na <sup>+</sup> is larger than Na atom
3	Mark the incorrect statement	A. Metallic character increase down the group B. Metallic character increase from left to right along a period C. Metallic character decrease from left to right along a period D. Metallic character remains the same down the group
4	Which one is an incomplete period	A. 4th B. 5th C. 6th D. 7th
5	Fluorine is in group VIIA of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. Iodine D. Sulphur
6	Which oxide is more basic in nature.	A. BeO B. MgO C. CaO D. BaO
7	In a period, from left to right in the periodic table, the size of atom generally.	A. Increases B. decreases C. Remains constant D. First increase upto the middle of period and then decreases
8	Transition elements are those	A. Which becomes before uranium B. Which becomes after uranium C. Which are prepared artificially D. Which are in between s-block and p - block elements
9	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F
10	Which one of the following sets has coinage metals is it.	A. Cu, Hg, Au B. Cu, Ag, Au C. Ag, Au, Hg D. Cu, Fe, Au
11	Which element when react with chlorine form polymeric halide.	A. Na B. Be C. Ba D. P
12	Keeping in view size of atoms ,which order is the correct one.	A. Mg > Sr B. Ba > Mg C. Lu > Ce D. Cl > I
13	Which one of the following elements burns in air to form an oxide which, when shaken with water, give a solution with a pH greater than 7.	A. Carbon B. Magnesium C. sulphur D. Hydrogen

14	Hydrogen resembles in properties with	<p>A. IA, IV A and VII A elements            B. III A, IV A and V A elements            C. II A, IV A and VI A elements            D. II A, III A and VII A elements</p>
15	Which one of the following elements burns in air to form an oxide which, when shaken with water, give a solution with a pH greater than 7.	<p>A. Carbon            B. Magnesium            C. sulphur            D. Hydrogen</p>
16	Which of the following the highest hydration energy	<p>A. <math>\text{Li}^{+}</math>            B. <math>\text{Na}^{+}</math>            C. <math>\text{K}^{+}</math>            D. <math>\text{Mg}^{2+}</math></p>
17	What are the total numbers of periods in the modern periodic table.	<p>A. 3            B. 5            C. 7            D. 8</p>
18	Which one of the following oxides is amphoteric in nature.	<p>A. MgO            B. Na<sub>2</sub>O            C. SO<sub>2</sub>            D. ZnO</p>
19	Which of the following has highest M.P	<p>A. Aluminium            B. Silicon            C. Phosphorus            D. Sulphur</p>
20	Which has greater hydration energy.	<p>A. Li<sup>+</sup>            B. Na<sup>+</sup>            C. K<sup>+</sup>            D. Mg<sup>2+</sup></p>