

Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Fluorine is in group VII A of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. lodine D. Sulphur
2	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F
3	In which group, melting point and boiling point increase downward in a group	A. IA B. II A C. VII A D. Both a and b
4	Fluorine is in group VIIA of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. lodine D. Sulphur
5	The basis of modern periodic table is	A. Electron affinity B. Atomic mass C. Ionization Potential D. Atomic number
6	In modern periodic table VI period contains elements	A. 8 B. 18 C. 10 D. 32
7	Number of elements in the first period of the periodic table is	A. 2 B. 8 C. 14 D. 18
8	Which hydride is intermediate in nature.	A. NaH B. BeH2 C. NH3 D. HCI
9	Which element has lowest melting point	A. Beryllium B. Magnesium C. Calcium D. Barium
10	Which of the following statement is correct	A. Na atom is smaller than Na ⁺ B. Na atom is larger than K atom C. F atom is smaller than F ⁻ D. F atom is larger than F ⁻
11	The element of 2nd period, which has highest ionization energy from the following is	A. Be B. C C. N D. O
12	Which element when react with chlorine form polymeric halide.	A. Na B. Be C. Ba D. P
13	What are the total numbers of periods in the modern periodic table	A. 3 B. 5 C. 7
14	Which one is an incomplete period	A. 4th B. 5th C. 6th D. 7th
		A. Alkali metals

15	Which class of elements shows law value of first ionization potential.	B. Alkaline earth metals C. Halogens D. Noble gases
16	Element of which group reacts with hydrogen and form ionic hydrides.	A. II A B. IV A C. V A D. VI A
17	Which of the following are alkaline earth metals?	A. Be, Mg, Ca B. Li, Na, K C. Fe, CO, Ni D. B, Al, Ga
18	The oxides of metal sare generally	A. Acidic B. Basic C. Neutral D. Amphoteric
19	Which one of the following elements has the largest second ionization energy.	A. O B. Na C. F D. Ne
20	In which block of periodic table non metals are present.	A. s B. p C. d D. f