

Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Which of the following the highest hydration energy	A. Li ⁺ B. Na ⁺ C. K ⁺ D. Mg ⁺⁺
2	What criteria did Mendeleev use in arranging his periodic table.	A. Atomic mass B. Atomic number C. Mass number D. Density
3	Keeping in view size of atoms ,which order is the correct one.	A. Mg > Sr B. Ba > Mg C. Lu > Ce D. Cl > I
4	In which block of periodic table non metals are present.	A. s B. p C. d D. f
5	Which is the longest periodic table	A. 4 B. 5 C. 6 D. 7
6	Which of the following are alkaline earth metals?	A. Be, Mg, Ca B. Li, Na, K C. Fe, CO, Ni D. B, Al, Ga
7	Which statement is incorrect.	A. All the metals are good conductors of electricity B. All the metals are good conductor of heat C. All the metal form positive ions D. All the metal form acidic oxides
8	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F
9	Which one of the following is a metalloid.	A. sulphur B. Antimony C. Mercury D. Zinc
10	Elements of Groups IIA are called	A. Alkali metals B. Alkaline earth metals C. Coinage metals D. Halogens
11	Which one of the following sets has coinage metals is it.	A. Cu, Hg, Au B. Cu, Ag, Au C. Ag, Au, Hg D. Cu, Fe, Au
12	What are the total numbers of periods in the modern periodic table.	A. 3 B. 5 C. 7 D. 8
13	Across a period from left to right in the periodic table, the melting and boiling point.	A. Decrease B. Increase C. Remain constant D. First increase upto the middle of period and then decrease
14	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
		A. Beo

15	Which oxide is more basic in nature.	B. MgO C. CaO D. BaO
16	In a period, from left to right in the periodic table, the size of atom generally.	A. Increases B. decreases C. Remains constant D. First increase upto the middle of period and then decreases
17	In which compound, oxidation state of sulphur is +6	A. H ₂ S B. H ₂ SO ₄ C. H ₂ SO ₃ D. SO ₃
18	Which has greater hydration energy.	A. Li ⁺ B. Na ⁺ C. K ⁺ D. Mg ⁺²
19	In which one of the following sets do all three particles have same number of total electrons	A. F ⁻ , Cl ⁻ , Br ⁻ B. Li ⁺ , Na ⁺ , K ⁺ C. N ^{<sup>-3,</sup>} , O ⁻² , F ⁻ , D. Na ⁺ , Mg ⁺² , K ⁺
20	The oxides of metals are generally	A. Acidic B. Basic C. Neutral D. Amphoteric