

## Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Electron affinity is measure of energy	A. Required to remover the electron B. Released by adding an electron C. Required to excite an electron D. Released by removing an electron
2	Which one of the following is a metalloid.	A. sulphur B. Antimony C. Mercury D. Zinc
3	Fluorine is in group VII A of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. lodine D. Sulphur
4	The concept of atomic number was introduced by	A. Alrazi B. Mendeleeve C. Moseley D. Dobereiner
5	lonic Hydrides react with water to form	A. Proton B. Hydride ions C. Hydroxide ions D. Hydronium ions
6	Which element has lowest melting point	A. Beryllium B. Magnesium C. Calcium D. Barium
7	In which block of periodic table non metals are present.	A. s B. p C. d D. f
8	Which one of the following sets has coinage metaals is it.	A. Cu, Hg, Au B. Cu, Ag, Au C. Ag, Au, Hg D. Cu, Fe, Au
9	What criteria did Mendeleev use in arranging his periodic table.	A. Atomic mass B. Atomic number C. Mass number D. Density
10	Which of the following has highest M.P	A. Aluminium B. Silicon C. Phosphorus D. Sulphur
11	In modern periodic table VI period contains elements	A. 8 B. 18 C. 10 D. 32
12	Transition elements are those	A. Which becomes before uranium B. Which becomes after uranium C. Which are prepared artificially D. Which are in between s-block and p - block elements
13	Which has greater hydration energy.	A. Li+ B. Na+ C. K+ D. Mg+2
14	The first ionization energy of Na, Mg, Al and Si are in theorder of.	A. Na < Mg < Al < Si B. Na > Mg> Al> Si C. Na > Mg< Al< Si D. Na< Mg> Al< Si
15	Mark the correct statement.	A. Na+ is smaller than Na atom B. Na+ is large than Na atom C. Cl- is smaller than Cl atom

		D. cl- and Cl are equal in size
16	Who gave the law of Triads in 1829?	A. Dobereiner B. Mosely C. Newland D. Mendeleev
17	Which one of the following elements burns in air to form an oxide which, when shaken with water, give sa solution with a pH greater than 7.	A. Carbon B. Magnesium C. sulphur D. Hydrogen
18	Element of which group reacts with hydrogen and form ionic hydrides.	A. II A B. IV A C. V A D. VI A
19	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
20	Elements of Groups IIA are called	A. Alkali metals B. Alkaline earth metals C. Coinage metals D. Halogens