

Chemistry Fsc Part 1 Chapter 11 Online Test

Sr	Questions	Answers Choice
1	When a chemical reaction is completed.	A. Instantaneous rate \neq average rate B. Instantaneous rate = average rate C. Instantaneous rate is zero D. Both average and instantaneous rates become zero
2	Which one of the following statements is incorrect.	A. Enzymes are protein in nature B. Enzymes are catalyst C. Enzymes can catalyze any reaction D. Urease is an enzyme
3	The substance which decrease the activity of a catalyst is called.	A. Promoter B. Activator C. Inhibitor D. Positive catalyst
4	A substance which itself is not a catalyst but increases the activity of a catalyst is called.	A. Promoter B. Poisoner C. Inhibitor D. Enzyme
5	A type of metals are usually used as catalyst.	A. Coinage metal B. Alkali metals C. Transition metals D. alkaline earth metals
6	Which statement is incorrect about activated complex	A. Short lived B. Maximum energy C. Unstable combination of atoms D. Less energy than E_a
7	When a reaction occurs in many steps then the slowest step is.	A. Mechanism step B. Rate determining step C. enthalpy determining step D. None of the above
8	Which statement is incorrect about order of reaction	A. It cannot be determined experimentally B. It is determined experimentally C. Sum of exponents in rate equation D. It can have fraction value
9	Photosynthesis a photochemical reaction has order of reaction	A. 0 B. 1 C. 2 D. Fractional order
10	When a chemical reaction is completed the	A. Instantaneous rate \neq average rate B. Instantaneous rate = average rate C. Instantaneous rate is zero D. Both average and instantaneous rates become zero