

Chemistry Fsc Part 1 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	What is teh oxidation state of sulfur in the sulfate ion SO ⁻² 4	A. +4 B. +6 C. +2 D. 0
2	Which set of the followig conditions results in higher ionization energy.	A. Smaller atom and greater nuclear charge B. Smaller atom and smaller nuclear charge C. Larger atom and greater nuclear charge D. Large atom and the smaller nuclear charge
3	Which one do you think is correct about metallic character.	A. It decrease frin tio ti vittin ub a griyo B. It increases from top to bottom in a group C. It remians constant fro left to right in a period D. It increases from left to right in a period
4	Which is the correct general trend in the variation of electron affinity in a group	A. It becomes less negative from top to bottom B. It becomes more negative from top to bottom C. It remains the same D. It has no definit trend and changes irregularly
5	Tendency to lose electrons called	A. metallic character B. Electron affinity C. Contained by the character of the character o
6	Metallic oxides arein nature.	A. Acidic B. Basic C. Amphotric D. Depends upon oxide
7	Noble gases are present in 18 group also called	A. I A B. II B C. VIII A D. VIII B
8	Peroxides are those in whoih oxidation state of oxygen is	A1 B. +2 D2 E1/2
9	Which of the following has highest ionization energy.	A. P B. S C. Ar D. Cl
10	2nd Value of electron affinity	A. Will be negaive always B. Will always be positive C. Equal to 1dy value D. Double than 1st Value
11	Which is the correct trend in variation of electrongativity along a period of the periodic table.	A. It tdecreases from left to right across a period B. It increases from left to right across a period C. It remians constant D. It has no definite trend
12	Group 16 elements are also called.	A. Chalocogens B. Pncitogens C. Alkali metals D. Haloges
		A. Anionic radius is generally smaller than atomic radius

A MgO B CO2 C Select basic oxide. B CO2 C Select cyps C D D D D D D D D	13	Which one is the correct statemetn among the following	b. Cationic radius is generally bigger than atomic radius c. Cationic radius is generally smaller than atomic radius D. Both anionic and cationic radius
Element that can show variable oxidation states. B. Ca C. Da C. Da	14	Select basic oxide.	B. CÖ2 C. SiO2
Modern periodic table provides information aboutelements. B. 110 C. 110 D. 130 A F B. Cl C. 14/p> D. 8. Cl C. P A. C. P A. D. B. Cl D. D. B. Na poNitrogen C. C. Cop>Carbon D. D. Cop>Carbon D. D. Cop>Carbon D. Cop>Carbon D. Cop>Carbon D. D. Cop>Carbon D. Cop>Carbon D. Cop>Carbon D. D. Cop>Carbon D. Cop>Cop>Carbon D. Cop>Cop>Cop>Cop>Cop>Cop>Cop>Cop>Cop>Co	15	Element that can show variable oxidation states.	B. Ca C. S
Highest electonegatie element is. B. C	16	Modern periodic table provides information aboutelements.	B. 118 C. 110
Select the element which has highest 1st ionization energy. B. Nitrogen C. C. Carbon D. Oxygen A. Golden yellow flame B. Golden brown flame C. Crimson red flame D. Intense white flame A. B. Elect which is not alkaline earth metal B. B. B. Corimson red flame D. Elect which is not alkaline earth metal B. B. B. B. B. Elect which is not alkaline earth metal D. C. Corimson red flame C. Corimson red flame A. B. B. Elect which is not alkaline earth metal B. Corimson red flame C. Corimson red flame C. Corimson red flame C. Corimson red flame D. D. Elect which is not alkaline earth metal	17	Highest electonegatie element is.	B. Cl C. I
19 Na burns with oxygen to give. B. Golden brown flame C. C. Crimson red flame D. Intense white flame A. Ba A. Ba B. Ba C. Crimson red flame A. Ba C. Crimson red flame A. Ba C. Select which is not alkaline earth metal C. C. Sr	18	Select the element which has highest 1st ionization energy.	B. Nitrogen C. Carbon
20 Select which is not alkaline earth metal B. B. BeC. Sr	19	Na burns with oxygen to give.	B. Golden brown flame C. Crimson red flame
	20	Select which is not alkaline earth metal	B. Be C. Sr