

Group Properties and Elements

Sr	Questions	Answers Choice
1	Metals oxides are .	A. Neutral B. Acidic in nature C. Basic in nature D. All of these
2	Non metallic oxides are.	A. Basic in nature B. Acidic in nature C. Amphoteric D. All of these
3	Non-metal form	A. Positive ion B. Neutral atom C. Negative ion D. None of these
4	Which group of elements has low ionization energies.	A. Alkaline earth metals B. Halogen C. Alkali metals D. Noble gases
5	Sodium metal has electrons.	A. 11 B. 10 C. 12 D. 14
6	Metals possess	A. Ionic bond B. Metallic bond C. Covalent bond D. Coordinate bond
7	Which are good conductor of heat and electricity.	A. Non metals B. Metalloids C. Metals D. All of these
8	Which metal exists in liquid form at room temperature.	A. Sodium B. Potassium C. Mercury D. None of these
9	Identify least reactive metals among the following	A. Copper B. Mercury C. Silver D. All of these
10	Which one is the most reactive metal.	A. Gold B. Sodium C. Potassium D. All of these
11	Metals are the elements which have	A. Electropositive character B. Electronegative character C. Both a and b D. None of the above
12	Non-metals are generally soft, but which one of the following is extremely hard.	A. Graphite B. Iodine C. Diamond D. Phosphorus