

## ECAT Chemistry MCQ's Test For Full Book

0	Overhann	Annual Chaire
Sr	Questions	Answers Choice
1	Standard reduction of Zn = -0.76 V and that of Ni is -0.25 V. On coupling them by a salt bridge which of these will act as anode	A. Salt bridge will act as anode     B. Zn will act as anode     C. Ni will act as anode     D. None of these
2	Question Image	A. Moles per dm <sup>3</sup> B. Partial pressures C. Number of moles D. Mole fractions
3	Zwitter ion is ion an amino acid	A. Polar B. Monopular C. Dipolar D. Non polar
4	Grignard reagent on reaction with a ketone forms	A. Tertiary alcohol B. Secondary alcohol C. Primary alcohol D. Carboxylic acid
5	Plastics are a pollution problem because many plastics	A. Are made from petroleum     B. Are very inflammable     C. Burn to produce toxic fumes     D. Decompose to produce toxic products
6	Which for the following has no dipole moment	A. HCI B. H <sub>2</sub> S C. H <sub>2</sub> O D. CO <sub>2</sub>
7	Question Image	A. The value of K <sub>p</sub> falls with a rise in temperature B. The value of K <sub>p</sub> falls with increasing pressure C. Adding V <sub>2</sub> O <sub>5</sub> catalyst increase the equilibrium yield of sulphur trioxide equilibrium yield of sulphur trioxide D. The value of K <sub>p</sub> is equal to K <sub>c</sub>
8	The other name for distribution law is	A. Dispersive law B. Partition law C. Avogadro's law D. separation law
9	HNO <sub>2</sub> acts as an/a	A. Acid B. Oxidizing agent C. Reducing agent D. All the three
10	Which of the following is a typical transition metal	A. Sc B. Y C. Ra D. Co
11	The radiations with wavelength shorter than violet light are called	A. Ultraviolet B. Infrared C. Microwave D. Radio frequency
12	Condensation of aldeydes withα-hydrogen gives:	A. Acetal B. Ketal C. Aldol D. Cannizzaro product
13	Replacement of Cl of Chlorobenzene to give phenol requires drastic conditions but chlorine of 2, 4-Dinitrochlorobenzene is readily replaced because	A. NO <sub>2</sub> makes the electron rich ring at ortho and para positions B. NO <sub>2</sub> withdraws electrons at metaposition C. NO <sub>2</sub> donate electrons at m-position D. NO <sub>2</sub> withdraws

## electrons at ortho and para position

19	Which of the following mixture of liquids show negative deviation form Raults law	A. Ethyl alcohol and ether B. HCl and water C. Phenol- water D. Chlorobenene-bromobenene
18	Which is a component of macronutrient?	A. Zn B. N C. Mo D. Cl
17	Which of the following gases is used for illuminating purpose?	A. Methane B. Ethane C. Propane D. Butane
16	In the ground state of an atom, the electron is present:	<ul><li>A. In the nucleus.</li><li>B. In the second shell.</li><li>C. Nearest to the nucleus.</li><li>D. farthest from the nucleus.</li></ul>
15	The correct order of 2nd I.P. of C,N,O and F is	A. O > F > N > C B. O > N > F > C C. C > N > O > F D. F > O > N > C
14	The detection of functional group is called	A. Numerical analysis     B. Qualitative analysis     C. Combustion analysis     D. Quantitative analysis