

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which one is not an argillaceous material.	A. Clay B. Shale C. State D. Gypsum
2	Select from the following the one which is Alcohol	A. $\text{CH}_{\substack{3}}\text{CH}_{\substack{2}}\text{OH}$ B. $\text{CH}_{\substack{3}}\text{OCH}_{\substack{3}}$ C. $\text{CH}_{\substack{3}}\text{COOH}$ D. $\text{CH}_{\substack{3}}\text{CH}_{\substack{2}}\text{Br}$
3	Which one of the following elements commonly exhibits oxidation states of +6 and +3 in aqueous solution.	A. Na B. Cr C. Mg D. C
4	The region of earth capable of supposing life is called.	A. Atmosphere B. Biosphere C. Dithosphere D. Hydrosphere
5	Biphenyl is	A. Monocyclic aromatic compound B. Polycyclic aromatic compound C. Polycyclic fused ring compound D. Alicyclic compound
6	The product of fermentation of sucrose is	A. Ethanol and H_2O B. Ethanol and CO C. Ethanol and CO_2 D. Glucose and CO_2
7	In group V-A elements the most electronegative elements is	A. Sb B. N C. P D. As
8	Absolute alcohol is that which is	A. 100% B. 95% C. Ethanol mixed with methanol D. Ethanol mixed with H_2O
9	According to Lewis concept ethers behave as	A. Acid B. Base C. Acid as well as a base D. None of them
10	Cyclic structure of benzene was proposed by	A. Dewar B. Faraday C. Donn D. Kekulé
11	Acetaldehyde can be prepared by the oxidation of.	A. Acetic acid B. Ethanol C. 1- Propanol D. Ethanonic acid
12	The conversion of ethanol to ethene is an example of.	A. Dehydration B. Hydration C. Hydrogenation D. Fermentation
13	Which substance is used to consulate rubber latex	A. Ethyl alcohol B. Acetaldehyde C. Acetic acid D. Water
14	In which substance phosphorus is not present.	A. Yolk of egg B. Bones C. Apatite D. Galena

15	What are the total numbers of periods in the modern periodic table.	B. 5 C. 7 D. 8
16	Which group when attached with benzene increases electron density of ring.	A. -COOH B. -NO C. -CHO D. -CH ₃
17	How much fresh water is used for domestic purpose	A. 8% B. 23% C. 69% D. 100%
18	Across a period from left to right in the periodic table, the melting and boiling point.	A. Decrease B. Increase C. Remain constant D. First increase upto the middle of period and then decrease
19	Which gas is evolved when CO ₂ is passed through KO ₂ .	A. O ₂ B. CO C. O ₃ D. None of these
20	When CO ₂ is made to react with ethyl magnesium iodide, followed by acid hydrolysis, the product formed is.	A. Propane B. Propanoic acid C. Propanal D. Propanol