

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which one of the following is not use of acetic acid.	A. Coagulant for rubber latex B. Local irritant C. Formation of rayon and silk D. Formation of alcohol
2	Which one of the following compound has octane number 100.	A. 2,2,4-trimethyl petane B. n- pentane C. 2,4-dimethyl pentane D. 2- methyl pentane
3	The pH of truly acidic rain is	A. 7-6.8 B. 6.5-6 C. 6-5.6 D. less than 5
4	Which one of the following gases exist in monoatomic form.	A. Ozone B. Nitrogen C. Krypton D. Phosphine
5	Which one of the following polymers is called a polyamide	A. Nylon B. Rayon C. Terylene D. Orlon
6	Acetic acid manufactured by	A. Distillation B. Fermentation C. Ozonolysis D. Esterification
7	The conversion of ethene to ethanol is an example of.	A. Hydration B. Dehydration C. Neutralization D. Esterification
8	Which one is perchloric acid	A. HClO B. HClO C. HClO ₃ D. HClO ₄
9	What is Glacial acetic acid	A. Pure acetic acid B. 95% acetic acid C. a mixture of acetic acid and glycerol D. Vinegar
10	Methyl alcohol can be distinguished from ethyl alcohol by	A. Action of Cl ₂ B. Action of NH ₃ C. Dissolving in H ₂ O D. NaOH + I ₂
11	The product of fermentation of sucrose is	A. Ethanol and H ₂ O B. Ethanol and CO C. Ethanol and CO ₂ D. Glucose and CO ₂
12	Which one of the following has highest melting and boiling points.	A. HF B. HBr C. HCl D. HI
13	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
14	Which class of elements shows low value of first ionization potential.	A. Alkali metals B. Alkaline earth metals C. Halogens D. Noble gases
15	In which process, an amino acid is produced	A. Wurtz synthesis B. Strecker synthesis C. Kolbe synthesis

D. Cannizzaro reaction

16	Boiling point range of petroleum ether.	<p>A. 5- 20 ^oC</p> <p>B. 10- 30 ^oC</p> <p>C. 20- 60 ^oC</p> <p>D. 30- 90 ^oC</p>
17	Which element reacts with alkali and H ₂ gas is produced.	<p>A. Be</p> <p>B. Mg</p> <p>C. Ca</p> <p>D. Sr</p>
18	The catalytic oxidation of methane produces	<p>A. CO + H₂O</p> <p>B. CO₂ + H₂O</p> <p>C. C₂ + H₂O</p> <p>D. H₃C - OH</p>
19	When carbon di oxide is bubbled through lime water, white precipitate is formed This precipitate is.	<p>A. COCl₂</p> <p>B. H₂CO₃</p> <p>C. CaO</p> <p>D. CaCO₃</p>
20	Transition elements are those	<p>A. Which becomes before uranium</p> <p>B. Which becomes after uranium</p> <p>C. Which are prepared artificially</p> <p>D. Which are in between s-block and p - block elements</p>