

Chemistry Fsc Part 2 Online Test

0-	Overstitute	Annuary Chaire
Sr	Questions	Answers Choice
1	Which statement is incorrect.	A. All the metals are good conductors of electricity B. All the metals are good conductor of heat C. All the metal form positive ions D. All the metal form acidic oxides
2	Which of the following is not a fatty acid	A. Propanoic acid B. Acetic acid C. Phthalic acid D. Butanoic acid
3	Which catalyst is used in contact process.	A. Fe2O3 B. V2O5 C. SO3 D. Ag2O
4	In ethene molecule how many carbon orbitals are equivalent and degenerate in nature.	A. 3 B. 4 C. 5 D. 6
5	Essential amino acids are those which	A. Are present in every protein B. Must be supplied to body through diet C. Contain two carbocyclic acid and one amino group D. Is syntherised by our body
6	Which one of the following is not a heterocyclic compound.	A. Furan B. Thiophene C. Pyridine D. Aniline
7	Which complex shows zero oxidation state of the transition metal.	A. [Fe(CO)5] B. K3[Fe(CN)6] C. K2[Fe(CN)6] D. [Cu(NH3)4]SO4
8	Which amino acid is present in cheese	A. Glycine B. Alanine C. Tyrosine D. Valine
9	A quality of raw water is improved by	A. Reduction B. Aeration C. Dehydration D. Incineration
10	Which one of the following alcohols will be formed when ethyl magnesium bromide reacts with acetone.	A. Primary alcohol B. Secondary alcohol C. Tertiary alcohol D. Dehydrin alcohol
11	When concentrated H2SO4 and solid sodium chloride react together at room temperature the product are.	A. Two salts only B. A salt and a base C. A salt and an acid D. A salt and water
12	The product of fermentation of sucrose is	A. Ethanol and H2O B. Ethanol and CO C. Ethanol and CO2 D. Glucose and CO2
13	During the manufacturing process of cement the temperature of the decomposition zone goes up to	A. 600 b style="color: rgb(34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C B. 800 b style="color: rgb(34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C C. 1000 b style="color: rgb(34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C D. 1200 b style="color: rgb(34, 34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C D. 1200 b style="color: rgb(34, 34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C D. 1200 b style="color: rgb(34, 34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C c. 1000 b style="color: rgb(34, 34, 34, 34); font-family: arial, sans-serif; font-size: 16px;">°C c. 1000 c. 1000 c. 1000 c. 1000 c. 1000

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14	Which allotrope of phosphorus has layers like graphite.	A. white phosphorus B. Red phosphorus C. Black Phosphorus D. Amorphous phosphorus
15	Acetic acid is manufactured by	A. distillation B. fermentation C. ozonalysis D. esterification
16	Which can produce ketone	A. Sec alcohol B. Calcium acetate C. Propyne D. All of these
17	Which of the following noble gas is used for arc welding and cutting	A. Helium B. Argon C. Xenon D. Radon
18	Methyl alcohol is not used	A. As a solvent B. As an anti freezing agent C. As a substitute for petrol D. For denaturing of ethyl alcohol
19	Which one of the following is not a nucleophile.	A. H2O B. H2S C. BF3 D. NH3
20	Which chemicals used for detection of amino acids.	A. NaOH + I2 B. Phenyl hydrazine C. Ninhydrin D. Benedict's solution