

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which one of the following compound is not of a polymer.	A. Starch B. Glucose C. Protein D. Nylon -6, 6
2	Which one is cyclic amino acid	A. Glycine B. Alanine C. Proline D. Aspartic acid
3	Which halogen occurs naturally in a positive oxidation state.	A. Fluorine B. Chlorine C. Bromine D. lodine
4	Which compound shows maximum hydrogen bonding with water.	A. CH3OH B. C2H5OH C. CH3-O-CH3 D. C6H5OH
5	How much does of methanol can cause death	A. 10-15 ml B. 15-20 ml C. 100- 250 ml D. has no effect
6	The mineral CaSO ₄ .2H ₂ O has the general name	A. Dolomite B. Gypsum C. Calcite D. Epsom Salt
7	Which one of the following proteins transports oxygen in blood stream.	A. Insulin B. Albumin C. Hemoglobin D. Globulin
8	Aromatic hydrocarbons are the derivatives of	A. Normal series of paraffins B. Alkene C. Benzene D. Cyclohexane
9	During manufacturing of cement , the temperature of pre heating zone is.	A. 500 ^o C B. 900 ^o C C. 1000 ^o C D. 1500 ^o C
10	When hydrolyzed, protein yields.	A. Fatty acid B. Glycerol C. Amino acid D. Nucleosides
11	Which is not a calcarious material	A. Clay B. Limestone C. Marble D. Chalk
12	In which compound, oxidation state of sulphur is +6	A. H2S B. H2SO4 C. H2SO3 D. SO3
13	Thickness of atmosphere is about how much kilometer above the surface of warth	A. 100 km B. 1000 km C. 10,000 km D. unlimited
14	To prevent corrosion, Iron pipes carrying drinking water are covered with zinc by	A. alley formation B. Electroplating C. Galvanizing D. Soldering
15	Which one is symmetrical ketone	A. Acetone B. Methyl ethyl ketone C. Methyl n propyl ketone D. 2- pentonone

16	The conversion of ethene to ethanol is an example of.	A. Hydration B. Dehydration C. Neutralization D. Esterification
17	Aluminum reacts with nitrogen to form	A. AIN B. Al ₂ N C. Al ₂ N ₃ D. Al ₄ N ₆
18	Which one of the following compounds show geometrical isomerism in it.	A. 1- pentene B. 1,1 dichloro ethane C. all of these D. 2- Pentene
19	The composition of white lead is	A. PbCO3.Ph(OH)2 B. PbCO3.2Pb(OH)2 C. 2PbCO3.Pb(OH)2 D. PbCO3
20	is not a nucleophile	A. H ₂ O B. NO ₃ C. BF ₃ D. NH ₃