

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Ecosystem is smaller units of.	A. Lithosphere B. Hydrosphere C. Atmosphere D. Biosphere
2	Which of the following are mono-saccharides	A. Fructose B. Sucrose C. Stach D. Cellulose
3	The process used to improve quality of gasoline	A. Thermal Cracking B. Reforming C. Combination D. Steam Cracking
4	P2O5 is usually used as	A. Drying agent only B. Reducing agent C. Both drying and reducing agent D. Both drying agent and oxidizing agent.
5	What is the function of Head box in paper making machine.	A. It reduces the thickness of paper B. It dry the paper C. It discharge teh pulp at the screen of fourdrinier table D. Web structure is consolidated
6	In IUPAC system, the name of K4[Fe(CN)6] is	A. Potassium ferricynaide B. Potassium ferrocynide C. Potassium Hexacyanoferrate (II) D. Poatssium hexacaynoferrate (III)
7	The concept of atomic number was introduced by	A. Alrazi B. Mendeleeve C. Moseley D. Dobereiner
8	Which of the following enzymes brings about the hydrolysis of fats	A. Urease B. Maltase C. Zymase D. Lipase
9	Which one is fatty acid	A. Benzoic acid B. Malonic acid C. Phtalic acid D. Palmitic acid
10	Which one of the following is not a heterocyclic compound.	A. Furan B. Thiophene C. Pyridine D. Aniline
11	Tincal is a mineral of	A. Al B. C C. Si D. B
12	Which is not an ore of iron	A. haematite B. Magnetite C. limonite D. Cassiterite
13	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
14	Lowest oxidation state of nitrogen is present in.	A. NH3 B. NO2 C. NO D. HNO3
15	Kaolin is a mineral of	A. Carbon B. Magnesium C. Silicon

		D. Aluminium
16	Elements of Groups IIA are called	A. Alkali metals B. Alkaline earth metals C. Coinage metals D. Halogens
17	Which compound show hydrogen bonding	A. C ₂ H ₆ B. C ₂ H ₅ Cl C. CH ₃ OCH ₃ D. C ₂ H ₅ OH
18	Which one of the following elements burns in air to form an oxide which, when shaken with water, give a solution with a pH greater than 7.	A. Carbon B. Magnesium C. sulphur D. Hydrogen
19	Which is most difficult to be oxidized	A. CH ₃ CHO B. CH ₃ COCH ₃ C. HCHO D. C ₂ H ₅ CHO
20	According to Lewis concept ethers behave as	A. Acid B. Base C. Acid as well as a base D. Noe of them