

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	In ethene molecule, the number of atoms which are present in the same plane are.	A. 2 B. 6 C. 3 D. 4
2	Which substance is water repellant and used in ceramic insulators.	A. Asbestos B. Lead compounds C. Silicon carbide D. silicones
3	When methane reacts with Cl2 is commonly known as	A. Mustard gas B. Laughing gas C. Phosgene gas D. Bio gas
4	The presence of a double bond in a compound is the sign of.	A. Saturation B. Unsaturation C. Substitution D. None of these
5	Methyl alcohol can be distinguished from ethyl alcohol by	A. Action of Cl2 B. Action of NH3 C. Dissolving in H2O D. NaOH +I2
6	In which substance phosphorus is not present.	A. Yolk of egg B. Bones C. Apatite D. Galena
7	Usually the percentage of moisture is paper is	A. 1- 3% B. 4-6% C. 6-8% D. 5%
8	Grignard's reagent is reactive due to	A. the presence of halogen atom B. the presence of Mg atom C. the polarity of C-Mg bond D. none of the above
9	Which is not a calcarious material.	A. Lime B. Clay C. Marble D. Marine shell
10	Which noble gas is used in radiotherapy	A. Neon B. Argon C. Krypta D. Radon
11	Cannizzaro's reaction is given by	A. Acetaldehyde B. Formaldehyde C. Propanal D. Propanone
12	The word alkali is derived from which language	A. Arabic B. Greek C. French D. German
13	Laughing gas is chemically	A. NO B. NO ₂ C. N ₂ O D. N ₄ O
14	In which of following processes are small organic molecules made into macromolecules	A. the cracking of petroleum fractions B. the fractional distillation of crude oil C. the polymerization of ethene D. the hydrolysis of proteins
15	Water is disinfected by a substance to avoid toxification	A. KMnO ₄ B. Alums C. O ₃ D. Cl ₂

16	Which substance is used in photography	A. AgCl B. AgBr C. AgI D. Ag3PO4
17	During the manufacturing process of cement the temperature of the decomposition zone goes up to.	A. 600 ^o C B. 900 ^o C C. 1000 ^o C D. 1200 ^o C
18	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
19	Acetic acid form a dimer in liquid phase because.	A. Low ionization constant of acid B. High solubility in water C. Hydrogen bonding D. Greater polarity
20	Colour of K2Cr2O7 is	A. Red B. Orange C. Green D. Yellow