

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Total number of d-block elements are	A. 10 B. 20 C. 30 D. 40
2	How many zones through which the charge passes in a rotary kiln	A. 4 B. 3 C. 2 D. 5
3	Which of these polymers is a synthetic polymer	A. Animal fat B. Starch C. Cellulose D. polyester
4	Oxide of Alkaline earth metal are	A. Acid B. Basic C. Neutral D. Amphoteric
5	Which one of the following substance does not react with Na.	A. CH ₃ COOH B. CH ₃ OCH ₃ C. CH ₃ OH D. C ₂ H ₅ OH
6	Structure of benzene is	A. Tetrahedral B. Trigonal planar C. Hexagonal planar D. Linear
7	Mark the correct statement.	A. Na ⁺ is smaller than Na atom B. Na ⁺ is large than Na atom C. Cl ⁻ is smaller than Cl atom D. cl ⁻ and Cl are equal in size
8	Oxime is an addition product obtained when aldehydes react with.	A. HCN B. NH ₂ OH C. Phenyl hydrazine D. H ₂ O
9	Which is not a calcareous material	A. Clay B. Limestone C. Marble D. Chalk
10	Gold dissolves in "Aqua Regia" due to formation of Halide. Point out correct halide	A. AuF ₃ B. AuCl ₃ C. AuBr ₃ D. AuI ₃
11	To which class of organic compounds soap belongs.	A. Ether B. Ketones C. Aldehyde D. Salt of an organic acid
12	Co-ordination number of Pt in Pt Cl(NO ₂)(NH ₃) ₄	A. 2- B. 4 C. 1 D. 6
13	The composition of white lead is	A. PbCO ₃ .Ph(OH) ₂ B. PbCO ₃ .2Pb(OH) ₂ C. 2PbCO ₃ .Pb(OH) ₂ D. PbCO ₃
14	In Down's cell anode is made up of.	A. Iron B. Copper C. Graphite D. Platinum
15	One of following is argillaceous material	A. Marble B. Clay C. Lime D. Marine Shell

16	Prilling is a process in which	A. Concentration of urea is increased B. Water is removed from urea C. Molten urea is converted to solid drop D. Some other ingredients are added to urea
17	Woody raw material for paper pulp is obtained from	A. Cotton B. Bagasse C. Poplar D. Rice straw
18	Which one of the following is inorganic fertilizer.	A. Manure B. Urea C. Ammonium nitrate D. All of these
19	Which one of the following gases is used for artificial ripening of fruits	A. Ethene B. Ethyne C. Methane D. Propane
20	Which has greater hydration energy.	A. Li+ B. Na+ C. K+ D. Mg+2