Fsc Part 1 Chemistry MCQ's Test For Full Book

Sr	Questions	Answers Choice
1	Which of the following has maximum heat capacity	A. Water B. Mercury C. Ethanol D. Iron
2	The space orientation for f-subshell is.	A. 3 B. 7 C. 1 D. 5
3	Overall order is sum of	A. Coefficients B. Exponents in rate law C. Moles D. Products
4	A flat curve in rate vs time gaph indicate.	A. Reaction completed B. High rate C. Constant rate D. Temperature increase
5	Concept of Triads was given by	A. Mendeleev B. L. Meyer C. Dobereiner D. Newlands
6	A polymer used for non -stick cookware is.	A. Teflon B. Nylon C. Plyvinyl chloride D. Polythene
7	Electrolysis of CuSO4 using copper eletrodes results in	A. Increase in eletolyte concentration B. Decrease in Cu2+ concentration C. No change in electrolyte composition D. Formation of new compound
8	What type of reation has a negative Delta H and Delta S?	A. Always spntaneous B. Never spontaneous C. Spontaneous at low temp D. Spontaneous at high temp
9	NaOH is not considered an Bronsted Lowry acid because.	A. It's neutral B. It's not soluble C. Doesn't donate H+ D. Doesn't produce H+
10	32 g of oxygen gas contains.	A. 6.02 x 10²³ moles B. 6.02 x 10²³ atoms C. 12.4 x 10²³ molecules D. 24.4 x 10²³ atoms
11	Subshells in an atom are filled with electrons in an increasing order of their energy values called.	A. aufbau pricniple B. Hund's rule C. Pauli's Exclusion Principle D. None
12	The cathode in an electrolytic cell is.	A. Positive electrode B. Negative electrode C. Neutral elecrode D. None
13	Which scientist first time observed eh periodiciy in the elements.	A. Dobereiner B. D. Mendeleev C. j. Newlands D. L. Meyer
14	In electroplating the object to be plated is connected to	A. Anode B. Cathode C. Both electordes

		D. Electrolyte
15	In an energy profile, peak represents.	A. Reactants B. Activated complex C. Products D. Catalyst
16	Which one of the following is not the applicatio of solubility products.	A. Determination of solubility from Ksp B. Common ion effect C. Predicting precipitation D. Dissociation of bonds
17	Untis of Kc depend on	A. Catalyst B. Reaction stoichimometty C. Activation energy D. Delta H
18	Acidic oxides are	A. Metallic oxides B. Non Metallic oxides C. Metalloid oxides D. All
19	Which of the following statemetns about ideal gases is true.	A. they have strong intermolecular forces B. Their particles have significant volume C. Their volume is mainly due to particle size D. They have negligible intermolecular forces
20	Surface tension usually with interasig temperature.	A. Increase B. Decrease C. Remain constant D. Doubles