

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Which of the following statement is not correct about isotopes.	A. they have same atomic number B. They have same number of protons C. They have same physical properties D. They have same chemical propeties
2	Without stoichiometry which industry cannot exist.	A. Meta B. Petroleum C. Leather D. Chemical
3	Formation of NO is	A. Exothrmic B. Endothermic C. No Heat Change D. None of these
4	During the formation of ionic bond heat is.	A. Remains same B. Absorbed C. Released D. Both a and b
5	Which element is the most reactive element?	A. Florine B. Oxygen C. Chlorine D. Nitrogen
6	Formation of ammonia from Nitrogen and hydrogen is an.	A. Exothermic reaction B. Endothermic reaction C. Both a and b D. No heat change
7	How many elements are present in each 2nd and 3rd period.	A. 2 B. 32 C. 18 D. 8
8	Atomic number of lanthanum is	A. 57 B. 58 C. 59 D. 60
9	The man made polymer is	A. Polystyrene B. Starch C. Protein D. Cellulose
10	Whcih one of the following is the weakest force among the atoms.	A. Intermolecular force B. Ionic force C. Metallic force D. Covalent forces
11	Which branch of chemistry helps to protect water that has been poisoned by soil.	A. Organic chemistry B. inorganic chemistry C. Environmental chemistry D. Geochemistry
12	Which of the following groups contain alkaline earth metals.	A. I A B. II A C. VII A D. VIII A
13	The reaction in which the products can recombine to form reactants are called.	A. Reversible Reaction B. Irreversible reactions C. Decomposition reactions D. Addition reactions
14	Which of the following elements is present in 1st period.	A. Hydrogen B. Helium C. Both a and b D. None of these
		A. KCl

15	Which one of the following will show negligible effect of temperature on its solubility.	<p>B. KNO_3 C. NaCl D. NaNO_3</p>
16	In an irreversible reaction equilibrium is.	<p>A. The forward reaction will be favoured B. No effect on forward or backward reaction C. No effect on backward reaction D. The backward reaction will be favoured</p>
17	Metals can form ions carrying charges.	<p>A. Uni positive B. Di positive C. Tri positive D. All of these</p>
18	The force of attraction between water molecules is.	<p>A. Ionic bonding B. Covalent bonding C. Hydrogen Bonding D. Coordinate Covalent bonding</p>
19	Formula of Ozone.	<p>A. O_2 B. S_8 C. CO_2 D. O_3</p>
20	Which one of the following halogens has lowest electronegativity	<p>A. Iodine B. Chlorine C. Fluorine D. Bromine</p>