

## 9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Uni of electron affinity is.	A. k J mol <sup>-1</sup> B. k jmol C. pm D. Newton
2	17th Group of elements contain electrons in their outer most shell	A. 4 B. 5 C. 7 D. 6
3	Which element has the highest melting point.	A. K B. Cs C. Na D. Rb
4	Which one of the following compounds will have the highest percentage of the mass of nitrogen?	A. N <sub>2</sub> H <sub>4</sub> B. CO (NH <sub>2</sub> ) <sub>2</sub> C. NH <sub>3</sub> D. NH <sub>2</sub> OH
5	Matter is present in neon signs in the state of	A. Supercritical fluid B. Gas C. Liquid Crystal D. Plasma
6	Which is released in anaerobic respiration.	A. Stearic acid B. Citric acid C. Lactic acid D. Amino Acid
7	Which group of the periodic table has the tendency to gain electrons.	A. Group -1 B. Group -17 C. Group-2 D. Group -18
8	Which metal has the lowest melting point?	A. Li B. Na C. Rb D. K
9	How many periods are present in the modern periodic table.	A. 7 B. 8 C. 10 D. 12
10	When we move from left to right in a period, atomic number	A. Decreases B. Increases C. First increases then decreases D. None of the above
11	The enthalpy of reaction C+O <sub>2</sub> → CO <sub>2</sub>	A. -571.6 kJ B. -393.5 kJ C. +53.8 kJ D. -110.5 kJ
12	How many elements are present in 7th period.	A. 2 B. 8 C. 18 D. 23
13	Which acid is used in the preparation of soap.	A. Oxalic Acid B. Citric Acid C. Stearic Acid D. Tartaric Acid
14	Washing clothes at 140 °F uses almost the energy as at 140 °C wash	A. Half B. Thrice C. Twice D. None of the above
15	Dative covalent bond is also known as	A. Covalent bond B. Ionic Bond C. Metallic Bond D. Coordinate covalent bond

16	The percentage of $^{238}\text{U}_{92}$ found in nature.	A. 97% B. 0.72% C. 98% D. 1.5%
17	Number of period in the periodic table are.	A. 7 B. 8 C. 5 D. 16
18	While baking which gas is responsible for raising the bread and making it soft?	A. Oxygen B. Carbon monoxide C. Carbon di oxide D. Nitrogen
19	In which isotope of oxygen there are the equal numbr of protons electons and neutorns.	A. $\text{O}^{16}$ B. $^{17}\text{O}$ C. $^{18}\text{O}$ D. None of these
20	Main group elements are arranged in .....groups.	A. 7 B. 6 C. 8 D. 10