

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Atmospheric region found between 50-58 km from the earth's.	A. Thermosphere B. Stratosphere C. Mesosphere D. Thermosphere
2	Dilute acid react with carbonates the produce product except _____	A. Salt B. Water C. Carbon di oxide D. Hydrogen gases
3	A substance which can behave as an acid as well as a base is called.	A. Acid B. Base C. Amphoteric D. Neutral
4	Iron and steel structure are damaged by.	A. Carbon monoxide B. Sulphur dioxide C. Methane D. Carbon dioxide
5	Chemical formula of matte will:	A. $Cu_2S.FeS$ B. Cu_2S C. Cu_2O D. FeS
6	Amino acid are the building blocks of .	A. Proteins B. Carbohydrates C. Vitamins D. fats
7	Which hydrocarbon decolorizes the pink colour of acidic solution of potassium permanganate?	A. CH_4 B. C_2H_4 C. C_2H_6 D. C_2H_5
8	If $Q_c < K_c$ the reaction goes in:	A. Forward B. Reverse C. At equilibrium state D. None
9	Methyl butanoate smells like;	A. a pine apple B. an apple C. an orange D. a lemon
10	In smelting process a small amount of coke is required because:	A. it is highly endothermic process B. it is highly exothermic process C. it is very fast process D. it is very slow process
11	Which protect our muscles from cramping?	A. Carbohydrats B. Vitamins C. Lipids D. Proteins
12	Natural source of citric acid is:	A. Rancid butter B. Fats C. Lemon D. Sour milk
13	The heat capacity of water is.	A. $4.2 Jg^{-1}K^{-1}$ B. $2.1 Jg^{-1}K^{-1}$ C. $3.2 Jg^{-1}K^{-1}$ D. $5.9 Jg^{-1}K^{-1}$
14	The most important oligosaccharide is:	A. Sucrose B. Glucose C. Fructose D. Maltose

15	If a liquid has a pH of 7 then it must.	<p>A. Be a colourless and odourless liquid</p> <p>B. Freeze at 0 ^oC and boil at 100 ^oC</p> <p>C. Be natural</p> <p>D. Be a solution containing water</p>
16	What is the pOH of Ca(OH) ₂ , 0.02 M Solution.	<p>A. 1.698</p> <p>B. 1.397</p> <p>C. 12.31</p> <p>D. 12.61</p>
17	Which one of the following is not a fossil fuel?	<p>A. Coal</p> <p>B. Natural gas</p> <p>C. Bio gas</p> <p>D. Petroleum</p>
18	Temporary hardness can be removed by_____.	<p>A. Quick lime</p> <p>B. Slaked lime</p> <p>C. Lime stone</p> <p>D. HCl</p>
19	Which ion cause water hardness?	<p>A. Al²⁺</p> <p>B. Mg²⁺</p> <p>C. Fe²⁺</p> <p>D. Na²⁺</p>
20	Ethyne is oxidized by alkaline KMnO ₄ then hydroxyl group add to the triple bond.	<p>A. Two</p> <p>B. Three</p> <p>C. Four</p> <p>D. Five</p>