

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Which ion cause water hardness?	A. Al^{2+} B. Mg^{2+} C. Fe^{2+} D. Na^{2+}
2	How many percentage blister copper is pure?	A. 95% B. 75% C. 98% D. 100%
3	Which one of the following ion does not cause hardness in water?	A. Ca^{+2} B. Mg^{2-} C. SO^{2-} D. Na^{+}
4	How many % age of nitrogen present in air by volume?	A. 70% B. 75% C. 78% D. 80%
5	Law of Mass Action was put forward by:	A. G.N.Lewis B. Lowry C. Arrhenius D. Gulbberg and Waage
6	Water dissolves Non-ionic compounds by:	A. Ion-Ion forces B. Dipole forces C. Dipole - Dipole forces D. Hydrogen Bonding
7	Which one of the following does not contain starch	A. Sugar cane B. Maize C. Barley D. Potatoes
8	Acid means	A. Sour B. Bitter C. Sweet D. Salty
9	A base is a substance which neutralizes an acid which of these substances is not a base?	A. Aqueous ammonia B. Sodium chloride C. Sodium Hydroxide D. Calcium Hydroxide
10	Which one of the following diseases causes liver inflammation?	A. Typhoid B. Jaundice C. Cholera D. Hapatitis
11	Which of one of the following is used as as fumigant:	A. Ethyl alcohol B. ethylene oxide C. ethylene glycol D. diethyl ether
12	Deficiency of Vitamin E causes.	A. rickets B. Scurvy C. Anemia in babies D. Night blindness.
13	Permanent hardness is removed by adding.	A. Na^{2+} zeolite. B. Soda lime C. Lime water D. Quick lime
14	DNA was discovered by	A. J .Watson B. Funk C. Robert brown D. Hopkins
15	Which salt is used as a table salt?	A. NaCl B. $\text{Na}^{2+}\text{CO}^{3-}$ C. $\text{Na}^{2+}\text{SiO}^{3-}$ D. NaCl

16	Which one of the mineral acid.	A. HCl B. H_2SO_4 C. HNO_3 D. All of these
17	Froth flotation process is used to concentrate.	A. Copper ore B. Iron ore C. Chromium ore D. Aluminum ore
18	Main component of natural gas is	A. Methane B. Propane C. Butane D. Propene
19	Acid rain affects the aquatic life by clogging fish gills because of:	A. Lead metal B. Chromium metal C. Mercury metal D. Aluminium metal
20	Ozone is an allotropic form of	A. Carbon B. Oxygen C. Sulphur D. Phosphorous