

## 9th Class Chemistry English Medium Online Test For Full Book

| Sr | Questions   | Answers Choice   |
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| 1  | Which one of the diseases causes severe diarrhea and can be fatal?                              | A. Jaundice<br>B. Dysentery<br>C. Cholera<br>D. Typhoid  |
| 2  | Which factor determines the severity of a pollutant?  | A. Chemical nature<br>B. concentration<br>C. Persistence<br>D. All of them   |
| 3  | Which are called paraffines?  | A. Alkyle<br>B. Alkynes<br>C. Alkenes<br>D. Alkanes  |
| 4  | Proteins are polymers of.   | A. Polysaccharide<br>B. Oligosaccharids<br>C. Amino acid<br>D. Nucleic acid  |
| 5  | Which one of the following does not contain proteins.   | A. In pulses<br>B. In Potatoes<br>C. In fruits<br>D. In eggs   |
| 6  | Smell of rancid butter is due to the presence of.   | A. Propanoic acid<br>B. Butanoic acid<br>C. Acitic acid<br>D. Citric acid  |
| 7  | Which of the following gas is used to destroy harmful bacteria in water.                        | A. Iodine<br>B. Chlorine<br>C. fluorine<br>D. Bromine  |
| 8  | Which acid is found in sour milk?   | A. `Formic acid<br>B. Lactic acid<br>C. Citric acid<br>D. Butric acid  |
| 9  | Forest fires and burning of wood emit.  | A. $\text{CO}_2$<br>B. $\text{NO}_2$<br>C. $\text{SO}_2$<br>D. $\text{Cl}_2$   |
| 10 | Dehalogenation of tetrahalides produces acetylene. This reaction takes place in the presence of | A. Sodium metal<br>B. Zinc metal<br>C. Magnesium metal<br>D. Potassium metal   |
| 11 | The heat capacity of water is greater than rocks.   | A. 4 times<br>B. 5 times<br>C. 6 times<br>D. 7 times   |
| 12 | The meaning of latin word acidus is.  | A. Sweet<br>B. Tasteless<br>C. Salty<br>D. Sour  |
| 13 | Number of vitamins in vitamin B complex is.   | A. 10<br>B. 8<br>C. 6<br>D. 12   |
| 14 | At dynamic equilibrium:   | A. The reactions stops to proceed<br>B. The amounts of reactants and products are equal<br>C. The speed of the forward is reverse reactions are equal<br>D. The reaction can no longer be reversed |
|    |   | A. $\text{C}_n(\text{H}_{2n})$   |

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| 15 | General formula of carbohydrate is.   | $C_6H_{12}O_6$<br>B. CH<br>C. $C_nH_{2n}$<br>D. $C_6H_{12}O_2$                  |
| 16 | The product of any Lewis acid-base reaction is a single specie called:                              | A. free radical<br>B. adduct<br>C. molecular ion<br>D. conjugate                |
| 17 | Lavoisier named binary compounds of oxygen acids in .   | A. 1787<br>B. 1790<br>C. 1815<br>D. 1828  |
| 18 | How many atmospheric temperature increase every year due to accumulation of CO <sub>2</sub> in air. | A. 0.03 °C<br>B. 0.05 °C<br>C. 1 °C<br>D. 2 °C                                  |
| 19 | Which acid is used for food preservation?   | A. Hydrochloric acid<br>B. Benzoic acid<br>C. Sulphuric acid<br>D. Nitric acid. |
| 20 | When HCl and KOH is reacted the salt formed is.   | A. Acidic<br>B. Basic<br>C. Normal<br>D. Complex                                |