

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	When the reaction causes to produced it is called.	A. Chemical equilibrium state B. Static equilibrium C. Dynamic equilibriun D. All
2	The coal in which the percentage of carbon is 60%	A. Peat B. Lignite C. Bituminous D. Anthracit
3	Formula of acetaldehyde is:	
4	There are _____ types of salts.	A. 2 B. 3 C. 6 D. 8
5	Metallurgy involves which of the following steps?	A. Mining and enrichment B. Reduction C. Refining and casting. D. All of these
6	Ozone is beneficial for us as it	A. Absorb infrared radiation. B. Absorb Ultraviolet radiations. C. Absorb Chlorofluorocarbons D. Absorb Air pollutant
7	The reaction goes from left to right , if:	A. $Q_c = K_c$ B. $Q_c > K_c$ C. $Q_c < K_c$ D. $Q_c = 0$
8	The carbon contents in lignite is.	A. 80% B. 50% C. 60% D. 70%
9	Monosaccharide consists of number of carbon atoms.	A. 2 to 4 B. 4 to 8 C. 3 to 9 D. 5 to 10
10	A substance which can behave as an acid as well as a base is called.	A. Acid B. Base C. Amphoteric D. Neutral
11	Chemical formula of matte will:	A. $Cu_2S.FeS$ B. Cu_2S C. Cu_2O D. FeS
12	Depending upon temperature variation, atmosphere ins divided into how many regions?	A. 1 B. 2 C. 3 D. 4
13	Which acid is present in our stomach.	A. Nitric acid B. Hydrochloric acid C. Sulphuric acid D. All of these
14	Which one is the simplest sugar which can not be hydrolyzed?	A. Glucose B. Sucrose C. Starch D. Cellulose
15	Density of water at 4°C is:	A. 0.976 $g\text{cm}^{-3}$ B. 1 $g\text{cm}^{-3}$ C. 0.956 $g\text{cm}^{-3}$ D. 1.1 $g\text{cm}^{-3}$
16	Ionic compounds are soluble in water due to:	A. Hydrogen bonding B. ion-dipole forces C. Dipole-dipole forces

D. Dipole -induced dipole forces

17 Which disease cause liver inflammation?

- A. Typhoid
- B. Jaundice
- C. cholera
- D. Hepatitis

18 Ethen is prepared by heating a mixture of ethanol and excess of concentrated sulphuric acid at:

- A. 150°C
- B. 180°C
- C. 250°C - 300°C
- D. 400°C - 450°C

19 Lewis acid-base concept have the following characteristics except:

- A. formation of an adduct
- B. Formation of a co-ordinate covalent bound
- C. Donation and acceptance of an electron pair
- D. Donation and acceptance of a proton

20 General formula of Alkynes is.

- A. $C_{6n}H_{2n-2}$
- B. C_nH_{2n+2}
- C. $C_{11n}H_{2n-2}$
- D. C_nH_{2n-2}