

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	You want to dry a gas which one of the following salt you will use?	A. CaCl_2 B. NaCl C. CaO D. Na_2SiO_3
2	The acid which is called king of chemicals.	A. Sulphuric Acid B. Nitric Acid C. Hydrochloric acid D. Acetic acid
3	General formula of alkyl radical is	A. $\text{C}_n\text{H}_{2n+2}$ B. $\text{C}_n\text{H}_{2n-2}$ C. $\text{C}_n\text{H}_{2n+1}$ D. C_nH_{2n}
4	50% dextrose contains approximate amount of energy.	A. 250 calories B. 16 Calories C. 170 calories D. 120 calories
5	Ethyl butanoate smells like.	A. Apple B. Pine apple C. Lemon D. Melon
6	Uric acid is found in	A. Urine B. Fats C. Apple D. Grapes
7	Atmosphere has regions.	A. One B. Two C. Three D. Four
8	Ethylene is present in natural gas sometimes to the extent of .	A. 10% B. 20% C. 30% D. 40%
9	Permanent hardness is removed by adding.	A. $\text{Na}_2\text{Zeolite}$. B. Soda lime C. Lime water D. Quick lime
10	In which of the following groups , oxygen is attached on both sides with carbon atoms.	A. Ketone group B. Ether group C. Aldehyde group D. Ester group
11	Which salt cause permanent hardness in water is:	A. $\text{Ca}(\text{HCO}_3)_2$ B. $\text{Mg}(\text{HCO}_3)_2$ C. CaCl_2 D. KCl
12	Water of crystallization is responsible for.	A. Melting points of crystals B. Boiling point of crystal C. Shape of crystal D. Transition points of crystals
13	When the magnitude of K_c is very large it indicates.	A. Reaction never goes to completion B. Reaction is in equilibrium state C. Reaction will complete after some time D. Reaction has almost to completion
14	The process of roasting during metallurgy of copper is carried out in a special furnace called.	A. Blast furnace B. Fire furnace C. Bessemer converter D. Reverberatory furnace

15	Which one of the following are amorphous solids?	A. Monosacchrides B. Oligosaccharides C. Polysaccharides D. All of them
16	Smell of rancid butter is due to the presence of.	A. Propanoic acid B. Butanoic acid C. Acitic acid D. Citric acid
17	Guldberg and waage put forward law of mass action in:	A. 1860 B. 1869 C. 1870 D. 1879
18	Which element do not causes toxicity in water?	A. Lead B. Arsenic C. Sodium D. Mercury
19	At dynamic equilibrium:	A. The reactions stops to proceed B. The amounts of reactants and products are equal C. The speed of the forward is reverse reactions are equal D. The reaction can no longer be reversed
20	Which pollutant is responsible for changing weather?	A. Air pollutant B. Water pollutant C. Soil pollutant D. All of them