

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Which one of the following is non- reducing in nature;	A. glucose B. fructose C. starch D. sucrose
2	Oxidation of alkenes produce.	A. Glyoxal B. Glycol C. Oxalic acid D. Formic acid
3	How many percentage to water contains oceans?	A. 67% B. 77% C. 87% D. 97%
4	Acetic acid is used for:	A. Elching designs B. Clearing metals C. Flavouring food D. Making explosives
5	When a reaction ceases to proceed further , it is called.	A. Chemical states B. Static state C. Physical state D. Dynamic equilibrium state
6	Blonde hair contains compounds of:	A. molybdenum B. iron C. titanium D. copper
7	Which is not an air pollutant?	A. CO ₂ B. CO C. SO ₂ D. NH ₃
8	Ozone id beneficial for us as it:	A. absorbs infrared radiations B. absorbs ultraviolet radiation C. absorbs chlorofluorocarbons D. absorbs air pollutants
9	The color of hydrogen Iodide is:	A. Colourless B. Black C. Red D. Pink
10	Alkanes are least reactive compounds because they are.	A. Saturated hydrocarbons. B. Unsaturated hydrocarbons. C. Both a and b D. None of the above.
11	Plants produces by the photosynthesis process.	A. Glucose B. Fructose C. Sucrose D. Maltose
12	Which one of the following are amorphous solids?	A. Monosacchrides B. Oligosaccharides C. Polysaccharides D. All of them
13	Permanent hardness can be removed by using .	A. Soda lime B. Sodium Zeolite C. Quick lime D. Lime water
14	_____ is the king of chemicals.	A. Hydrochloric acid B. Nitric Acid C. Sulphuric acid D. Phosphoric acid
15	Pitch is black residue of	A. Coke B. Coal-tar C. Coal D. Coal gas

16	Acid means	A. Sour B. Bitter C. Sweet D. Salty
17	Acid occurring in sour milk	A. Citric acid B. Lactic acid C. Butyric acid D. Malic acid
18	A pollutant is a waste material that pollutes.	A. Air B. Water C. Soil D. All of them
19	Monosaccharide consists of number of carbon atoms.	A. 2 to 4 B. 4 to 8 C. 3 to 9 D. 5 to 10
20	The reaction between an acid and a base produces.	A. Salts and water B. Salt and gas C. Salt and an acid D. Salt and an base
