

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	Potassium ferrocyanide salt is.	A. Normal B. complex C. Basic D. Acidic
2	99% of atmosphere consists of .	A. N_2 and H_2 B. N_2 and O_2 C. N_2 and CO_2 D. O_2 and CO_2
3	Which one of the following is polymerized to form benzene?	A. ethane B. ethene C. methane D. ethyne
4	When Bronsted and Lowry put forward acid base concept?	A. 1913 B. 1923 C. 1933 D. 1943
5	Which of the following is soluble in water.	A. Organic acids B. Glucose C. Alcohol D. All of them
6	Alkalis react with ammonium salt to liberate.	A. SO_2 B. CO_2 C. NH_3 D. H_2
7	conjugate base of HCl is:	A. HCl^{sup-} B. CH C. Cl D. NH_3
8	A salt always	A. Contains ions B. Contains water of crystallization C. Dissolves in water D. Forms crystals which conduct electricity
9	Which gas is responsible in warming the atmosphere?	A. Nitrogen B. Hydrogen C. Helium D. Fluorine
10	What is the percentage of carbon in anthracite?	A. 60% B. 70% C. 80% D. 90%
11	the chemical formula of chalcopryite is:	A. Cu_2S B. $CuFe_2S$ C. CuS D. FeS
12	Calcium carbonate is in water.	A. Insoluble B. Sparingly C. None of them D. Soluble
13	Which hydrocarbon molecule would have no effect on the aqueous solution of bromine?	A. CH_4 B. $C_{10}H_{20}$ C. C_2H_4 D. C_2H_2
14	The lives of aquatic plants and animals are indirectly related to concentration of dissolved gas in water.	A. Nitrogen B. Hydrogen C. Oxygen D. Carbon

15	Swimming pools are cleaned by the process.	A. Chlorination B. Hydrogenations C. Saponification D. None of them
16	Which disease cause when humans use water of industrial effluents.	A. Cancer B. Asthma C. Jaundice D. cholera
17	Which one of the following does not contain proteins.	A. In pulses B. In Potatoes C. In fruits D. In eggs
18	Which one of the following is Lewis base?	A. AlCl_3 B. H^+ C. NH_3 D. BF_3
19	_____ is synthetic fiber.	A. Cotton B. Woal C. Nylon D. Silk
20	Ozone id beneficial for us as it:	A. absorbs infrared radiations B. absorbs ultraviolet radiation C. absorbs chlorofluorocarbons D. absorbs air pollutants