

9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	The nature of enzyme is.	A. Protein B. Vitamin C. Fats D. Carbohydrate
2	Formula of Decane is.	A. $C_{10}H_{20}$ B. $C_{10}H_{22}$ C. $C_{10}H_4$ D. $C_{10}H_{16}$
3	Maltose is commonly found in .	A. Cereal B. Milk C. Cotton D. Honey
4	How many % age of urea is used as fertilizers?	A. 80% B. 90% C. 95% D. 98%
5	Which one of the following is a fat soluble vitamin?	A. A B. E C. K D. All of these
6	Permanent hardness can be removed by using .	A. Soda lime B. Sodium Zeolite C. Quick lime D. Lime water
7	Which one of the following does not contain starch	A. Sugar cane B. Maize C. Barley D. Potatoes
8	Which one of the following organic compound is found in gasoline?	A. C_2H_4 B. C_3H_8 C. C_7H_{10} D. $C_{12}H_{26}$
9	Acid used for cleaning metals , tanning and in printing industries, is:	A. H_2SO_4 B. HNO_3 C. CH_3COOH D. HCl
10	Mg^{2+} and Ca^{2+} ions react with soap to form calcium and magnesium salts of fatty acids called.	A. Gelatin B. Scum C. Paste D. None of them
11	Which one of the following is used as jet fuel.	A. Kerosene oil B. Lubricating oil C. Fuel oil D. Diesel oil
12	Dilute acid react with carbonates the produce product except _____	A. Salt B. Water C. Carbon di oxide D. Hydrogen gases
13	A disease is caused by excess of bile pigments in the blood is.	A. Typhoid B. Jaundice C. Cholera D. Dysentery
14	A base is a substance which neutralizes an acid which of these substances is not a base?	A. Aqueous ammonia B. Sodium chloride C. Sodium Hydroxide D. Calcium Hydroxide
15	All acids turn blue litmus.	A. Red B. Blue C. Pink D. White

16	Molecular formula of butyne is .	A. C_4H_6 B. C_3H_4 C. C_4H_7 D. C_4H_8
17	Rancid butter has	A. Foul smell B. Rotten egg smell C. Pungent smell D. No smell
18	High concentration of which metal clogs the fish gills.	A. Zinc B. Aluminium C. Sodium D. Copper
19	Which gas is also known as life gas for plants.	A. CO B. CO_2 C. O_2 D. NO_2
20	Which acid is found in Ant sting?	A. Citric Acid B. Formic Acid C. Uric Acid D. Sulphuric acid