

## 9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	How many percentage to water contains oceans?	A. 67% B. 77% C. 87% D. 97%
2	Carbon black is used in the manufacture of.	A. Dry cleaning B. Shoe polishes. C. Fertilizer D. None of these
3	Carbonization process is the conversion of	A. Coal into coal gas B. Coal into wood C. Wood into coal D. Wood into coal tar
4	Which one is not an Arrhenius base?	A. NaOH B. KOH C. $\text{Ca}(\text{OH})_2$ D. $\text{NH}_3$
5	Number of vitamins in vitamin B complex is.	A. 10 B. 8 C. 6 D. 12
6	Potassium ferrocyanide salt is.	A. Normal B. complex C. Basic D. Acidic
7	Which factor determines the severity of a pollutant?	A. Chemical nature B. concentration C. Persistence D. All of them
8	Which type of reactions take place in both directions?	A. Decomposition reactions B. Irreversible reactions C. Reversible reactions D. Addition reactions
9	Uric Acid is present in	A. Apple B. Fats C. Urine D. Grapes
10	General formula of alkynes is:	A. $\text{C}_n\text{H}_n$ B. $\text{C}_n\text{H}_{2n}$ C. $\text{C}_n\text{H}_{2n+2}$ D. $\text{C}_n\text{H}_{2n-2}$
11	Cause of Night blindness is.	A. Deficiency of vitamin B B. Deficiency of vitamin A C. Deficiency of vitamin D D. Deficiency of vitamin E
12	Which one contains double covalent bond?	A. Pentane B. Ethylene C. Acetylene D. All
13	The order of reactivity of hydrogen halides with alkenes is.	A. $\text{HI} > \text{HBr}$ B. $\text{HBr} > \text{HI}$ C. $\text{HCl} > \text{HBr}$ D. $\text{HBr} > \text{HCl}$
14	Sea water is unfit for drinking purpose due to the presence of.	A. Salts B. Algae C. Fishes D. All of them
15	Formula of Potassium ferrocyanide.	A. $\text{K}_4[\text{Fe}(\text{CN})_6]$ B. $\text{K}_3[\text{Fe}(\text{CN})_4]$ C. $\text{K}_4[\text{Fe}(\text{CN})_5]$ D. $\text{K}_3[\text{Fe}(\text{CN})_6]$

		<p>C. <math>K_{c2}[\text{Fe}(\text{CN})_6]^{4-}</math></p> <p>D. <math>K_{c4}[\text{Fe}(\text{CN})_5]^{4-}</math></p>
16	Which gas is used to destroy harmful bacteria in water?	<p>A. Iodine</p> <p>B. Chlorine</p> <p>C. Flourine</p> <p>D. Bromine</p>
17	Which one of the following organic compound is found is gasoline?	<p>A. <math>\text{C}_2\text{H}_4</math></p> <p>B. <math>\text{C}_3\text{H}_8</math></p> <p>C. <math>\text{C}_7\text{H}_{10}</math></p> <p>D. <math>\text{C}_{12}\text{H}_{26}</math></p>
18	Reactions which have comparable amounts of reactants and products at equilibrium state have:	<p>A. very small <math>K_c</math> value</p> <p>B. Very large <math>K_c</math> value</p> <p>C. Moderate <math>K_c</math> value</p> <p>D. None of these</p>
19	According to Lewis concept a base is a substance which can donate.	<p>A. Proton</p> <p>B. Electron pair</p> <p>C. Neutron</p> <p>D. Electron</p>
20	Hypoglycemia result due to:	<p>A. high cholesterol level</p> <p>B. low cholesterol level</p> <p>C. low sugar level in body</p> <p>D. high sugar level in body</p>