

## 9th Class Chemistry English Medium Online Test For Full Book

Sr	Questions	Answers Choice
1	The % age of carbon in coal is	A. 40-60 B. 50-70 C. 40-80 D. 40-90
2	According to Bronsted and Lowry concept a base is a substance that can accept.	A. Proton B. Electron pair C. Neutron D. Electron
3	Which gas is produced during ripening of bananas.	A. Methane B. Ethene C. Acetylene D. Carbon dioxide
4	Which of the following salt is soluble in water.	A. NaCl B. KCl C. $\text{Na}_2\text{SO}_4$ D. All of them
5	What is the percentage of carbon in anthracite?	A. 60% B. 70% C. 80% D. 90%
6	Which one contains double covalent bond?	A. Pentane B. Ethylene C. Acetylene D. All
7	The plant use	A. Carbon dioxide B. Oxygen C. Nitrogen D. Sulphur
8	Which is not a character of $\text{SO}_2$	A. it is a colourless gas B. It has irritating smell C. It causes suffocation D. It does not form sulphuric acid
9	Which one of the following is a fat soluble vitamin?	A. A B. E C. K D. All of these
10	Which vitamin is soluble in water?	A. Vitamin A B. Vitamin E C. Vitamin D D. Vitamin C
11	The bond in the product of Lewis acid base reaction.	A. Ionic B. Covalent C. Metallic D. Coordinate covalent
12	Which are called paraffines?	A. Alkyls B. Alkynes C. Alkenes D. Alkanes
13	At which temperature does water show maximum density.	A. 0 °C B. 100 °C C. 4 °C D. -4 °C
14	Acute lead poisoning causes dysfunction of.	A. Kidney B. Liver C. CNS D. All of them
15	Olefins is a Latin word meaning:	A. inert B. less reactive C. oil forming D. most reactive

16	A salt always	<p>A. Contain ions B. Contains water of crystallization C. Dissolves in water D. Forms crystals which conduct electricity</p>
17	A reverse reaction is that.	<p>A. Which proceed from left to high B. In which reactants reacts to form products C. Which slow down gradually D. Which speed up gradually</p>
18	The Earth's atmosphere is getting hotter because of.	<p>A. Increasing concentration of CO B. Increasing concentration of CO<sub>2</sub> C. Increasing concentration of O<sub>3</sub> D. Increasing concentration of SO<sub>2</sub></p>
19	The value of K <sub>c</sub> depends on.	<p>A. Temperature B. Pressure C. Volume D. Atmosphere</p>
20	Acidic solutions have pH value.	<p>A. Less than 7 B. Greater than 7 C. equal to 7 D. None of these</p>