

## General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Chemical bond teh forms between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these
2	Oxygen is a	A. Monoalent element B. Divalent element C. Trivalent element D. Tertravalent element
3	Which of teh following is cation.	A. H <sub>2</sub> B. H <sub>2</sub> O C. CO <sub>2</sub> D. NH <sub>4</sub> <sup>+</sup>
4	How many atoms orm one molecuel of chlorine.	A. 1 B. 2 C. 3 D. 4
5	Chemical formula of Sulphuric Acid.	A. HCl B. H <sub>2</sub> SO <sub>4</sub> C. HNO <sub>3</sub> D. CH <sub>3</sub> COOH
6	Chemical bond in O <sub>2</sub> is.	A. Ionic B. Covalent C. Metallic D. None of these
7	While writing a chemical formula positive ionis written on.	A. Right B. Left C. Center D. All of these
8	Positively charged atom or grouping of atoms is called.	A. Anion B. Cation C. Neutron D. Proton
9	Chemical bond between N <sub>2</sub> Molecule is	A. Ionic B. Single covalent C. Double covalent D. Triple vocalent
10	Which of the following is an anion.	A. F <sup>-</sup> B. Cl <sup>-</sup> C. S <sup>-2</sup> D. NH <sub>3</sub>
11	Types of coalent bond are	A. 1 B. 2 C. 3 D. 4
12	Outermost shell of sodium atom has ..... electrons.	A. 1 B. 2 C. 3 D. 4
13	Chemical bond is O <sub>2</sub> is.	A. Ionic B. Covalent C. Metallic D. None of these
14	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
15	Negatively charged atom or grop of atoms is called.	A. Proton B. Electron C. Anion D. Cation

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16 An atm with stable elecrronic configuration can exist.

A. Reacts readily  
B. Can not exist  
C. Independently  
D. Combined form

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17 Which of the following has stble elecronic configuration.

A. H  
B. He  
C. C  
D. Li

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18 Outermost shell of chlorine stom has..... electrons.

A. 1  
B. 2  
C. 6  
D. 7

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19 When water molecule gain a proton it form

A. H<sup>+</sup>  
B. OH<sup>-</sup>  
C. H<sub>3</sub>O  
D. `<blockquote style="margin: 0 0 0 40px; border: none; padding: 0px;">H2O</blockquote>`

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20 Whcih of the following has double covalent bound.

A. Br<sub>2</sub>  
B. CO<sub>2</sub>  
C. HCl  
D. N<sub>2</sub>

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