

General Science 7th Class English Medium Chapter 6 Online Test

| Sr | Questions | Answers Choice |
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| 1 | Outermost shell of sodium atom has electrons. | A. 1 B. 2 C. 3 D. 4 |
| 2 | Which of the following has double covalent bond. | A. Br ₂ B. CO ₂ C. HCl D. N ₂ |
| 3 | Positively charged atom or grouping of atoms is called. | A. Anion B. Cation C. Neutron D. Proton |
| 4 | When water molecule gain a proton it form | A. H ⁺ B. OH ⁻ C. H ₃ O ⁺ D. <div style="border: 1px solid black; padding: 2px; display: inline-block;">H₂O</div> |
| 5 | Valency of nitrogen is | A. 1 B. 2 C. 3 D. 4 |
| 6 | Chemical bond forms between metallic and non-metallic atoms. | A. Covalent B. Ionic C. Metallic D. None of these |
| 7 | Oxygen is a | A. Monovalent element B. Divalent element C. Trivalent element D. Tetravalent element |
| 8 | Which of the following has stable electronic configuration. | A. H B. He C. C D. Li |
| 9 | Negatively charged atom or group of atoms is called. | A. Proton B. Electron C. Anion D. Cation |
| 10 | Which of the following is cation. | A. H ₂ B. H ₂ O C. CO ₂ D. NH ₄ ⁺ |
| 11 | An atom with stable electronic configuration can exist. | A. Reacts readily B. Can not exist C. Independently D. Combined form |
| 12 | Outermost shell of chlorine atom has..... electrons. | A. 1 B. 2 C. 6 D. 7 |
| 13 | While writing a chemical formula positive ion is written on. | A. Right B. Left C. Center D. All of these |
| 14 | A molecule consists of two or more | A. Compounds B. Atoms C. Formulae D. Element |
| 15 | Which of the following is an anion. | A. F B. Cl C. S ²⁻ D. |

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| | | D. NH ₃ |
| 16 | Chemical bond in O ₂ is. | A. Ionic B. Covalent C. Metallic D. None of these |
| 17 | How many atoms form one molecule of chlorine. | A. 1 B. 2 C. 3 D. 4 |
| 18 | Chemical bond between N ₂ molecule is | A. Ionic B. Single covalent C. Double covalent D. Triple covalent |
| 19 | Types of covalent bond are | A. 1 B. 2 C. 3 D. 4 |
| 20 | Chemical formula of Sulphuric Acid. | A. HCl B. H ₂ SO ₄ C. HNO ₃ D. CH ₃ COOH |