

## General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Outermost shell of sodium atom has ..... electrons.	A. 1 B. 2 C. 3 D. 4
2	Whcih of the following has double covalent bound.	A. Br <sub>2</sub> B. CO <sub>2</sub> C. HCl D. N <sub>2</sub>
3	Positively charged atom or grouping of atoms is called.	A. Anion B. Cation C. Neutron D. Proton
4	When water molecule gain a proton it form	A. H <sup>+</sup> B. OH <sup>-</sup> C. H <sub>3</sub> O D. <pre style="margin: 0 0 0 40px; border: none; padding: 0px;">H2O</pre>
5	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
6	Chemical bond teh forms between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metalic D. None of these
7	Oxygen is a	A. Monoalent element B. Divalent element C. Trivalent element D. Tertravalent element
8	Which of the following has stble elecronic configuration.	A. H B. He C. C D. Li
9	Negatively charged atom or grop of atoms is called.	A. Proton B. Electron C. Anion D. Cation
10	Which of teh following is cation.	A. H <sub>2</sub> B. H <sub>2</sub> O C. CO <sub>2</sub> D. NH <sub>4</sub> <sup>+</sup>
11	An atm with stable elecrronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
12	Outermost shell of chlorine stom has..... electrons.	A. 1 B. 2 C. 6 D. 7
13	While writing a chemical formula positive ionis written on.	A. Right B. Left C. Center D. All of these
14	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
15	Which of the following is an anion.	A. F B. Cl C. S <sup>-2</sup>

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16 Chemical bond in O<sub>2</sub> is.

A. Ionic  
B. Covalent  
C. Metallic  
D. None of these

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17 How many atoms form one molecule of chlorine.

A. 1  
B. 2  
C. 3  
D. 4

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18 Chemical bond between N<sub>2</sub> Molecule is

A. Ionic  
B. Single covalent  
C. Double covalent  
D. Triple covalent

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19 Types of covalent bond are

A. 1  
B. 2  
C. 3  
D. 4

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20 Chemical formula of Sulphuric Acid.

A. HCl  
B. H<sub>2</sub>SO<sub>4</sub>  
C. HNO<sub>3</sub>  
D. CH<sub>3</sub>COOH

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