

General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Chemical bond in O_2 is.	A. Ionic B. Covalent C. Metallic D. None of these
2	Negatively charged atom or group of atoms is called.	A. Proton B. Electron C. Anion D. Cation
3	Which of the following is an anion.	A. F B. Cl C. S^{2-} D. NH_3
4	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
5	Outermost shell of chlorine atom has..... electrons.	A. 1 B. 2 C. 6 D. 7
6	Types of covalent bond are	A. 1 B. 2 C. 3 D. 4
7	How many atoms form one molecule of chlorine.	A. 1 B. 2 C. 3 D. 4
8	Chemical bond in O_2 is.	A. Ionic B. Covalent C. Metallic D. None of these
9	Which of the following is a cation.	A. H_2 B. H_2O C. CO_2 D. NH_4^+
10	Outermost shell of sodium atom has electrons.	A. 1 B. 2 C. 3 D. 4
11	Chemical formula of Sulphuric Acid.	A. HCl B. H_2SO_4 C. HNO_3 D. CH_3COOH
12	Oxygen is a	A. Monovalent element B. Divalent element C. Trivalent element D. Tetravalent element
13	When water molecule gains a proton it forms	A. H^+ B. OH^- C. H_3O^+ D. H_2O
14	Which of the following has a double covalent bond.	A. Br_2 B. CO_2 C. HCl D. N_2
15	While writing a chemical formula, positive ions are written on	A. Right B. Left C. Center

		D. All of these
16	An atm with stable electronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
17	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
18	Chemical bond between N ₂ Molecule is	A. Ionic B. Single covalent C. Double covalent D. Triple vocalent
19	Which of the following has stable electronic configuration.	A. H B. He C. C D. Li
20	Chemical bond forms between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these