

General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	How many periods are there in the Periodic table.	A. Five B. Six C. Seven D. Eight
2	The maximum number of electrons that N-Shell accommodate are.	A. 2 B. 8 C. 18 D. 32
3	Formula showing maximum number of electrons in a shell.	A. n^2 B. $2n^2$ C. $3n^2$ D. $4n^2$
4	Atomic number of element present in period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 D. 7
5	The central part of an atom is called.	A. Electron B. Proton C. Neutron D. Nucleus
6	The mass of an atom is almost due to the.	A. Electron B. Proton C. Nucleus D. Neutrons
7	Positively charged particle of an atom is called.	A. Electron B. Proton C. Atom D. Neutron
8	The elements of group 18 are called.	A. Alkali metals B. Noble gases C. Alkaline metals D. Transition metals
9	Area around nucleus where electrons are found are called.	A. Orbit B. Shell C. Both a and b D. None of these
10	The particle of an atom having negative charge on it.	A. Electron B. Proton C. Neutron D. Nucleus
11	Element present in groups have same number of in their outermost shell.	A. Electron B. Proton C. Atoms D. Neutron
12	Letter K is used to express the shell around the nucleus of an atom.	A. 1st B. 2nd C. 3rd D. 4th
13	The charge present on atom is	A. Positive B. Negative C. Both a and b D. Neutral
14	The total number of electrons which M shell can accommodate	A. 2 B. 8 C. 18 D. 32
15	An element with fifteen protons in its atomic nucleus has atomic number.	A. 5 B. 10 C. 15 D. 20

16	The negatively charged particles around nucleus are called.	A. Electron B. Proton C. Atom D. Neutron
17	The particles of an element having different mass numbers.	A. Proton B. Neutrons C. Isotopes D. Atoms
18	There are groups in periodic table.	A. 2 B. 8 C. 18 D. 32
19 are fundamental particles of an atom.	A. Electron B. Proton C. Neutron D. All of these
20	The number of electrons present in an atom is equal to number of.	A. Nucleus B. Neutrons C. Proton D. Atom