

## General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	How many periods are there in the Periodic table.	A. Five B. Six <b>C. Seven</b> D. Eight
2	The maximum number of electrons that N-Shell accommodate are.	A. 2 B. 8 C. 18 <b>D. 32</b>
3	Formula showing maximum number of electrons in a shell.	A. $n^2$ <b>B. <math>2n^2</math></b> C. $3n^2$ D. $4n^2$
4	Atomic number of element present in period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 <b>D. 7</b>
5	The central part of an atom is called.	A. Electron B. Proton C. Neutron <b>D. Nucleus</b>
6	The mass of an atom is almost due to the.	A. Electron B. Proton <b>C. Nucleus</b> D. Neutrons
7	Positively charged particle of an atom is called.	A. Electron <b>B. Proton</b> C. Atom D. Neutron
8	The elements of group 18 are called.	A. Alkali metals <b>B. Noble gases</b> C. Alkaline metals D. Transition metals
9	Area around nucleus where electrons are found are called.	A. Orbit B. Shell <b>C. Both a and b</b> D. None of these
10	The particle of an atom having negative charge on it.	<b>A. Electron</b> B. Proton C. Neutron D. Nucleus
11	Element present in groups have same number of ..... in their outermost shell.	<b>A. Electron</b> B. Proton C. Atoms D. Neutron
12	Letter K is used to express the shell around the nucleus of an atom.	<b>A. 1st</b> B. 2nd C. 3rd D. 4th
13	The charge present on atom is	A. Positive B. Negative C. Both a and b <b>D. Neutral</b>
14	The total number of electrons which M shell can accommodate	A. 2 B. 8 <b>C. 18</b> D. 32
15	An element with fifteen protons in its atomic nucleus has atomic number.	A. 5 B. 10 <b>C. 15</b> D. 20

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16 The negatively charged particle around nucleus are called.  
A. Electron  
B. Proton  
C. Atom  
D. Neutron

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17 The particles of an element having different mass numbers.  
A. Proton  
B. Neutrons  
C. Isotopes  
D. Atoms

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18 There are ..... groups in periodic table.  
A. 2  
B. 8  
C. 18  
D. 32

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19 ----- are fundamental particles of an atom.  
A. Electron  
B. Proton  
C. Neutron  
D. All of these

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20 The number of electrons present in an atom is equal to number of.  
A. Nucleus  
B. Neutrons  
C. Proton  
D. Atom

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