

General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	Formula showing maximum number of electrons in a shell.	A. n^2 B. $2n^2$ C. $3n^2$ D. $4n^2$
2	There are groups in periodic table.	A. 2 B. 8 C. 18 D. 32
3	Element present in groups have same number of in their outermost shell.	A. Electron B. Proton C. Atoms D. Neutron
4	The charge present on atom is	A. Positive B. Negative C. Both a and b D. Neutral
5	Positively charged particle of an atom is called.	A. Electron B. Proton C. Atom D. Neutron
6	The total number of electrons which M shell can accommodate	A. 2 B. 8 C. 18 D. 32
7	The mass of an atom is almost due to the.	A. Electron B. Proton C. Nucleus D. Neutrons
8	In elements are arranged in order of increasing atomic numbers.	A. Group B. Period C. Orbit D. Shell
9	The particle of an atom having negative charge on it.	A. Electron B. Proton C. Neutron D. Nucleus
10	The particles of an element having different mass numbers.	A. Proton B. Neutrons C. Isotopes D. Atoms
11	Atomic number of element present in period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 D. 7
12	The elements of group 18 are called.	A. Alkali metals B. Noble gases C. Alkaline metals D. Transition metals
13	Area around nucleus where electrons are found are called.	A. Orbit B. Shell C. Both a and b D. None of these
14	Letter K is used to express the shell around the nucleus of an atom.	A. 1st B. 2nd C. 3rd D. 4th
15	An element with fifteen protons in its atomic nucleus has atomic number.	A. 5 B. 10 C. 15 D. 20

16	How many periods are there in the Periodic table.	A. Five B. Six C. Seven D. Eight
17	The negatively charged particles around the nucleus are called.	A. Electron B. Proton C. Atom D. Neutron
18	The maximum number of electrons that an N-Shell accommodate are.	A. 2 B. 8 C. 18 D. 32
19	The number of electrons present in an atom is equal to the number of.	A. Nucleus B. Neutrons C. Proton D. Atom
20	The central part of an atom is called.	A. Electron B. Proton C. Neutron D. Nucleus