

General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	Letter K is used to express the shell around the nucleus of an atom.	A. 1st B. 2nd C. 3rd D. 4th
2	The maximum number of electrons that N-Shell accommodate are.	A. 2 B. 8 C. 18 D. 32
3	In elements are arranged in order of increasing atomic numbers.	A. Group B. Period C. Orbit D. Shell
4	The particles of an element having different mass numbers.	A. Proton B. Neutrons C. Isotopes D. Atoms
5	Positively charged particle of an atom is called.	A. Electron B. Proton C. Atom D. Neutron
6	Element present in groups have same number of in their outermost shell.	A. Electron B. Proton C. Atoms D. Neutron
7	The particle of an atom having negative charge on it.	A. Electron B. Proton C. Neutron D. Nucleus
8	Atomic number of element present in period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 D. 7
9	The mass of an atom is almost due to the.	A. Electron B. Proton C. Nucleus D. Neutrons
10	The central part of an atom is called.	A. Electron B. Proton C. Neutron D. Nucleus
11	The charge present on atom is	A. Positive B. Negative C. Both a and b D. Neutral
12	The negatively charged particle around nucleus are called.	A. Electron B. Proton C. Atom D. Neutron
13	The elements of group 18 are called.	A. Alkali metals B. Noble gases C. Alkaline metals D. Transition metals
14	An element with fifteen protons in its atomic nucleus has atomic number.	A. 5 B. 10 C. 15 D. 20
15	The number of electrons present in an atom is equal to number of.	A. Nucleus B. Neutrons C. Proton D. Atom

16 The total number of electrons which M shell can accommodate
A. 2
B. 8
C. 18
D. 32

17 Formula showing maximum number of electrons in a shell.
A. n^2
B. $2n^2$
C. $3n^2$
D. $4n^2$

18 How many periods are there in the Periodic table.
A. Five
B. Six
C. Seven
D. Eight

19 ----- are fundamental particles of an atom.
A. Electron
B. Proton
C. Neutron
D. All of these

20 There are groups in periodic table.
A. 2
B. 8
C. 18
D. 32
