

MDCAT Chemistry Chapter 13 Fundamental principles of organic chemistry Online Test

Sr	Questions	Answers Choice
1	During the nitration of benzene the nitrating agent is	A. NO ₃ B. NO ₂ ⁺ C. NO ₂ ⁻ D. HNO ₃
2	A compound that has a nucleophilic carbon?	A. C ₂ H ₂ B. C ₂ H ₄ C. C ₃ H ₈ D. C ₆ H ₆
3	Which of the following tests helps to distinguish between alkyne and alkene?	A. Lucas test B. Tollen's reagent test C. Baeyer's test D. Fehling's solution test
4	Baeyer's reagent is mixture of	A. HCl & ZnCl B. Aqueous bromine C. Alkaline KMnO ₄ D. Mix of Br ₂ & KMnO ₄
5	The substitution of a 'H' by '-NO ₂ ' group in benzene is called	A. Nitration B. Sulphonation C. Ammonolysis D. Reduction of benzene
6	The origin of acidic nature of alkyne is?	A. small size of C B. Small size of H C. polarity of triple bond D. sp hybridization
7	Benzene cannot undergo the ----- directly	A. Substitution reaction B. Addition reaction C. Oxidation reaction D. Elimination reaction
8	When 1-butene reacts with bromine, the product formed will be	A. 1, 3-dihydroxy butane B. But-1, 2-diol C. 1, 3-dihydroxy butan-diol D. 1,2-dibromo butane
9	Ethane when completely halogenated in excess of chlorine can form	A. Hexachloroethane B. Dichloroethane C. Pentachloroethane D. 1.1.2.2-tetrachloroethane
10	Which of the following compound reacts slower than benzene in the electrophilic substitution.	A. Phenol B. Nitrobenzene C. Toluene D. Aniline
11	2-Propenol, on rearrangement, yields	A. Propanal B. Propanone C. 2-propanol D. Both A and B
12	Which of the following is not an electrophilic substitutional reaction of benzene?	A. Free radical chlorination of benzene B. Friedel Craft alkylation C. Sulphonation D. Nitration
13	Glyoxal molecule has?	A. two carbonyl groups B. One aldehydic and one carbonyl group C. Two aldehydic groups D. Two carboxyl group
14	Which group activates the benzene ring	A. -COOH B. -COR C. -CHO D. -OH
15	The reaction that generates an ionic bond is	A. Halogenation of ethene B. polymerization of ethene C. Hydrogenation of ethyne

D. Reaction of ethyne with sodamide

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| 16 | The addition of HCl to ethene gives? | A. Chloroethane
B. 1,2-dichloroethane
C. 1,1-dichloroethane
D. 2-chloroethane |
| 17 | 2,5-dimethyl-1-hexene has | A. Two sp ² hybridized carbons
B. Six sp ² hybrid carbons
C. Two double bonds
D. Four pi electrons |
| 18 | Among the following the polycyclic aromatic compound is | A. Styrene
B. Naphthalene
C. Toluene
D. Acetophenone |
| 19 | Naphthalene has two fused aromatic ring of carbon atom the molecular formula | A. C ₁₀ H ₈
B. C ₁₀ H ₁₄
C. C ₁₀ H ₁₀
D. C ₁₂ H ₁₂ |
| 20 | Hydration of ethene is an example of | A. Electrophilic addition
B. Electrophilic substitution
C. Nucleophilic addition
D. Nucleophilic substitution |