

Chemistry Fsc Part 2 Chapter 3 Online Test

Sr	Questions	Answers Choice
1	The composition of white lead is	A. $\text{PbCO}_3 \cdot \text{Ph(OH)}_2$ B. $\text{PbCO}_3 \cdot 2\text{Pb(OH)}_2$ C. $2\text{PbCO}_3 \cdot \text{Pb(OH)}_2$ D. PbCO_3
2	Boric acid cannot be used	A. As antiseptic in medicine B. For washing eyes C. In soda bottles D. For Enamals and Glazes
3	Tincal is a mineral of.	A. Al B. B C. Si D. C
4	Boric acid is used as eye wash due to its	A. Weakly acidic property B. Antiseptic nature C. Antibiotic nature D. Weakly basic properties
5	Aluminium oxide is.	A. Acidic oxide B. Basic oxide C. Amphoteric oxide D. None of these
6	Lead monoxide is	A. Amphoteric B. Neutral C. Acidid D. Basic
7	Boron usually exist in nature as	A. Borides B. Oxborates C. Free element D. It is a artificial element
8	Which one of following is used in cosmetics	A. Talc B. Asbestos C. Sodium sulphate D. Aluminium Sulphate
9	Aluminium does not corrode because.	A. It is a semi metal B. It is a silver shining metal C. It does not react with O_2 D. It forms a protective layer of Al_2O_3
10	Tincal is a mineral of	A. Al B. C C. Si D. B
11	Which is used in the leather industry	A. Borax B. Boric acid C. Boric oxide D. Tetra Boric acid
12	Kaolin is a mineral of	A. Carbon B. Magnesium C. Silicon D. Aluminium
13	Which are correct oxidation states for lead	A. +1, +2 B. +3, +5 C. +2, +4 D. +4 only
14	Which is not used of Borax	A. Softening of water B. As medicine for washing eyes C. As flux in metallurgical operations D. To make quartz
15	Which of the following elements is not present abundantly in earth's crust	A. Silicon B. Aluminium C. Sodium D. Oxygen

16 The chief of aluminum is

A. Na^3AlF_6
B. $\text{Al}^2\text{O}^3 \cdot 2\text{H}_2\text{O}$
C. $\text{Al}^2\text{O}^3 \cdot 3\text{H}_2\text{O}$
D. $\text{Al}^2\text{O}^3 \cdot 3\text{H}_2\text{O}$

17 Which one of the following elements exhibits s-inert pair effect.

A. B
B. Al
C. Pb
D. Sc

18 A solution of borax in water is.

A. Acidic
B. Alkaline
C. Neutral
D. None of these

19 In which compound, the oxidation state of boron is not +3

A. H_2BO_3
B. $\text{Ca}_2\text{B}_2\text{O}_11$
C. $\text{Na}_2\text{B}_4\text{O}_7$
D. Mg_3B_2

20 Ortho boric acid on strong heating gives.

A. Borax
B. Boron oxide
C. Metaboric acid
D. Tetraboric acid
