Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	The oxide of beryllium is	A. Acidic B. Basic C. Amphoteric D. None of these
2	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be
3	Oxide of Alkaline earth metal are	A. Acid B. Basic C. Neutral D. Amphoteric
4	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	A. Gain water and remains solid B. Gain water and becomes liquid C. Lose mass and remain solid D. Remain unchanged on heating.
5	Which is called milk of magnesia.	A. MgCO3 B. Mg(OH)2 C. MgSO4 D. MgCl2
6	Point out the element which forms super oxide	A. Li B. Na C. K D. C
7	When water is added to plaster of paris, it changes to a hard mass. Its volume also expands upto	A. 1% B. 2% C. 3% D. 5%
8	Dolomite is a carbonate of	A. Be B. Mg C. Na D. Ba
9	Which meal is least reactive	A. Li B. NaOH C. K D. Pb
10	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
11	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
12	Magnesium metal does not burn in the jar of	A. N2 B. O2 C. Ne D. N2 and O2
13	Nelson's cell is used to prepare	A. NaOH B. Na₂CO₃ C. Na metal D. NaCl
14	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
15	When carbon di oxide is bubbled through lime water, white precipitate is formed This precipitate is.	A. COCl2 B. H2CO3 C. CaO D. CaCO3

16	The Milk of magnesia is used for the treatment of	A. Basicity B. Rancidity C. Acidity D. Jaundance
17	---------- does not belong to Alkaline-Earth metals	A. Be B. Ra C. Ba D. Rn
18	Which on eof the following alkaline earth metal form peroxide when heated with oxygen at 600 ^{o}C	A. Magnesium B. Calcium C. Strontium D. Barium
19	Which ion will have the maximum value of heat of hydration.	A. Na+ B. Ca+ C. Ba+2 D. Mg+2
20	Ca Mg3 (SiO3)2 is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos