

Chemistry Fsc Part 2 Chapter 2 Online Test

Cr.	Questions	Anguara Chaica
Sr	Questions	Answers Choice
1	Which gas is evolved when CO2 is passed through KO2.	A. O2 B. CO C. O3 D. None of these
2	When carbon di oxide is bubbled through lime water, white precipitate is formed This precipitate is.	A. COCI2 B. H2CO3 C. CaO D. CaCO3
3	Which element reacts with alkali and H2 gas is produced.	A. Be B. Mg C. Ca D. Sr
4	Which of the following gas will turn lime water milky	A. Cl ₂ B. NO ₂ C. CO D. CO ₂
5	Which one does not belong to the alkaline earth metals	A. Be B. Ba C. Ra D. Rn
6	Which one of the following does not belong to alkaline earth metals.	A. Be B. Ra C. Ba D. Rn
7	Ca Mg3 (SiO3)2 is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos
8	Magnesium metal does not burn in the jar of	A. N2 B. O2 C. Ne D. N2 and O2
9	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
10	The oxides of beryllium are	A. Basicity B. Rancidity C. Amphoteric D. none of all
11	Which of the following compounds is used to make spray, which has fungicidal action.	A. BaSO4 B. Na2CO3 C. CaO D. CaSO4.2H2O
12	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be
13	Which elements are deposited at the cathode during the electrolysis of brine in diaphragm cell	A. H ₂ B. Ba C. Ra D. Rn
14	Which one of the following is not an alkali metal	A. Francium B. Caesuym C. Rubidium D. Radium
15	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	A. Gain water and remains solid B. Gain water and becomes liquid C. Lose mass and remain solid D. Remain unchanged on heating.

16	Which is insoluble in water.	A. BeSO4 B. MgSO4 C. CaSO4 D. BaSO4
17	The element cesium bears resemblance with	A. Ca B. Cr C. Both of the above D. None of the above
18	Which on eof the following alkaline earth metal form peroxide when heated with oxygen at 600 $^{\rm o}{\rm C}$	A. Magnesium B. Calcium C. Strontium D. Barium
19	In which ore carbon is present	A. Magnesite B. Epsom salt C. Barite D. Sylvite
20	Chile saltpeter has the chemical formula	A. NaNO3 B. KNO3 C. Na2B4O7 D. Na2CO3.H2O