

## Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	Which metal is least reactive	A. Li B. NaOH C. K D. Pb
2	In which ore carbon is present	A. Magnesite B. Epsom salt C. Barite D. Sylvite
3	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
4	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
5	Setting of Plaster of Paris is accompanied with	A. Hydrogen bonding B. Hydration C. Dehydration D. None of these
6	Down's cell is used to prepare.	A. Sodium carbonate B. Sodium bicarbonate C. sodium metal D. Sodium hydroxide
7	Which of the following gas will turn lime water milky	A. $\text{Cl}_2$ B. $\text{NO}_2$ C. CO D. $\text{CO}_2$
8	Point out the element which forms super oxide	A. Li B. Na C. K D. C
9	Which one of the following alkaline earth metal form peroxide when heated with oxygen at 600 °C	A. Magnesium B. Calcium C. Strontium D. Barium
10	Which one of the following has lowest first ionization energy.	A. Li B. Na C. Rb D. Cs
11	Which gas is evolved when $\text{CO}_2$ is passed through $\text{KO}_2$ .	A. $\text{O}_2$ B. CO C. $\text{O}_3$ D. None of these
12	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be
13	The most metallic elements from the following is	A. Arsenic B. Oxygen C. Antimony D. Bismuth
14	$\text{Ca Mg}_3 (\text{SiO}_3)_2$ is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos
15	Nelson's cell is used to prepare	A. NaOH B. $\text{Na}_2\text{CO}_3$ C. Na metal D. NaCl

16	Which ion will have the maximum value of heat of hydration.	A. $\text{Na}^+$ B. $\text{Ca}^+$ C. $\text{Ba}^{+2}$ D. $\text{Mg}^{+2}$
17	Which one of the following does not belong to alkaline earth metals.	A. Be B. Ra C. Ba D. Rn
18	Which of the following sulphates is not soluble in water.	A. Sodium sulphate B. Potassium sulphate C. Zinc sulphate D. Barium Sulphate
19	Which metal oxide is insoluble in water	A. $\text{MgO}$ B. $\text{CaO}$ C. $\text{SrO}$ D. $\text{BaO}$
20	Sodium is collected over molten sodium chloride in Down's cell, because.	A. It does not react with $\text{NaCl}$ B. It is lighter in weight and floats on molten $\text{NaCl}$ C. Its reactivity is greater than $\text{NaCl}$ D. None of these