

Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	Which is more basic	A. RbOH B. NaOH C. KOH D. Li OH
2	Which metal oxide is insoluble in water	A. MgO B. CaO C. SrO D. BaO
3	Which one of the following has lowest first ionization energy.	A. Li B. Na C. Rb D. Cs
4	Which one is not use of lime	A. Extraction of metals B. Ceramic industry C. Preparation of bleaching powder D. Preparation of gypsum
5	Which of the following gas will turn lime water milky	A. Cl_2 B. NO_2 C. CO D. CO_2
6	Setting of Plaster of Paris is accompanied with	A. Hydrogen bonding B. Hydration C. Dehydration D. None of these
7	Which substance is used to lower the melting point of NaCl in Down's cell	A. CaCO_3 B. CaCl_2 C. MgSO_4 D. AlCl_3
8	Oxide of Alkaline earth metal are	A. Acid B. Basic C. Neutral D. Amphoteric
9	Which one of the following does not belong to alkaline earth metals.	A. Be B. Ra C. Ba D. Rn
10	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	A. Gain water and remains solid B. Gain water and becomes liquid C. Lose mass and remain solid D. Remain unchanged on heating.
11	The word alkali is derived from which language	A. Arabic B. Greek C. French D. German
12	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral D. Amphoteric
13	Which hydroxide decomposes on heating.	A. LiOH B. NaOH C. KOH D. CaOH
14	Chile saltpeter has the chemical formal	A. NaNO_3 B. KNO_2 C. $\text{Na}_2\text{B}_4\text{O}_7$ D. $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$
15	Which is more basic	A. RbOH B. NaOH C. KOH D. LiOH

		D. Li OH
16	Magnesium metal does not burn in the jar of	A. N ₂ B. O ₂ C. Ne D. N ₂ and O ₂
17	The oxide of beryllium is	A. Acidic B. Basic C. Amphoteric D. None of these
18	Ca Mg ₃ (SiO ₃) ₂ is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos
19	Which is insoluble in water.	A. BeSO ₄ B. MgSO ₄ C. CaSO ₄ D. BaSO ₄
20	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be