

## Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	----- does not belong to Alkaline-Earth metals	A. Be B. Ra C. Ba D. Rn
2	Point out the element which forms super oxide	A. Li B. Na C. K D. C
3	Oxide of Alkaline earth metal are	A. Acid B. Basic C. Neutral D. Amphoteric
4	When carbon di oxide is bubbled through lime water, white precipitate is formed This precipitate is.	A. COCl <sub>2</sub> B. H <sub>2</sub> CO <sub>3</sub> C. CaO D. CaCO <sub>3</sub>
5	The most metallic elements from the following is	A. Arsenic B. Oxygen C. Antimony D. Bismuth
6	Which one of the following does not belong to alkaline earth metals.	A. Be B. Ra C. Ba D. Rn
7	Dolomite is a carbonate of	A. Be B. Mg C. Na D. Ba
8	Which one of the following does not belong to alkaline earth metals	A. Be B. Ra C. Ba D. Rn
9	Magnesium metal does not burn in the jar of	A. N <sub>2</sub> B. O <sub>2</sub> C. Ne D. N <sub>2</sub> and O <sub>2</sub>
10	Which one of the following is not an alkali metal	A. Francium B. Caesium C. Rubidium D. Radium
11	Which oxide is amphoteric in nature.	A. BeO B. MgO C. CaO D. BaO
12	Sulphur is essential constituent for plants. Which is not role of sulphur	A. Chlorophyll development B. Development of root system C. Constituents of some proteins D. Increase transpiration force
13	Which element is present in chlorophyll	A. Ca B. Mg C. K D. Be
14	Down's cell is used to prepare.	A. Sodium carbonate B. Sodium bicarbonate C. sodium metal D. Sodium hydroxide
15	Ca Mg <sub>3</sub> (SiO <sub>3</sub> ) <sub>2</sub> is the composition of.	A. Dolomite B. Gypsum C. Calcite D. Asbestos

16	The crystals of caustic soda, NaOH, are hygroscopic when these crystals are exposed to air.	<p>A. Gain water and remains solid</p> <p>B. Gain water and becomes liquid</p> <p>C. Lose mass and remain solid</p> <p>D. Remain unchanged on heating.</p>
17	Which one of the following elements has the same oxidation number in all its known compounds.	<p>A. Beryllium</p> <p>B. Nitrogen</p> <p>C. Sulphur</p> <p>D. Chlorine</p>
18	In Down's cell anode is made up of.	<p>A. Iron</p> <p>B. Copper</p> <p>C. Graphite</p> <p>D. Platinum</p>
19	Which metal oxide is insoluble in water	<p>A. MgO</p> <p>B. CaO</p> <p>C. SrO</p> <p>D. BaO</p>
20	Chile saltpeter has the chemical formula	<p>A. <math>\text{NaNO}_3</math></p> <p>B. <math>\text{KNO}_2</math></p> <p>C. <math>\text{Na}_2\text{B}_4\text{O}_7</math></p> <p>D. <math>\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}</math></p>