

Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Which of the following has highest M.P	A. Aluminium B. Silicon C. Phosphorus D. Sulphur
2	Across a period from left to right in the periodic table, the melting and boiling point.	A. Decrease B. Increase C. Remain constant D. First increase upto the middle of period and then decrease
3	Ionization energy of calcium is.	A. Lower than that of barium B. Lower than that of magnesium C. Higher than that of beryllium D. Lower than that of strontium
4	Which class of elements shows low value of first ionization potential.	A. Alkali metals B. Alkaline earth metals C. Halogens D. Noble gases
5	Who gave the law of Triads in 1829?	A. Dobereiner B. Mosely C. Newland D. Mendeleev
6	Which element has lowest melting point	A. Beryllium B. Magnesium C. Calcium D. Barium
7	Which hydride is intermediate in nature.	A. NaH B. BeH ₂ C. NH ₃ D. HCl
8	Which statement is incorrect.	A. All the metals are good conductors of electricity B. All the metals are good conductor of heat C. All the metal form positive ions D. All the metal form acidic oxides
9	Which is the longest periodic table	A. 4 B. 5 C. 6 D. 7
10	Mark the correct statement.	A. Na ⁺ is smaller than Na atom B. Na ⁺ is large than Na atom C. Cl ⁻ is smaller than Cl atom D. Cl ⁻ and Cl are equal in size
11	What are the total numbers of periods in the modern periodic table.	A. 3 B. 5 C. 7 D. 8
12	The element of 2nd period, which has highest ionization energy from the following is	A. Be B. C C. N D. O
13	In which block of periodic table non metals are present.	A. s B. p C. d D. f
14	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F
		A. Na atom is smaller than Na ⁺

15	Which of the following statement is correct	<p></sup> B. Na atom is larger than K atom C. F atom is smaller than F<sup>-</sup> </sup> D. F atom is larger than F<sup>-</sup> </sup></p>
16	Element of which group reacts with hydrogen and form ionic hydrides.	<p>A. II A B. IV A C. V A D. VI A</p>
17	Which one of the following elements burns in air to form an oxide which, when shaken with water, give a solution with a pH greater than 7.	<p>A. Carbon B. Magnesium C. sulphur D. Hydrogen</p>
18	In modern periodic table VI period contains elements	<p>A. 8 B. 18 C. 10 D. 32</p>
19	What criteria did Mendeleev use in arranging his periodic table.	<p>A. Atomic mass B. Atomic number C. Mass number D. Density</p>
20	Fluorine is in group VIIA of periodic table. Its chemistry will most closely resembles that of.	<p>A. Argon B. Boron C. Iodine D. Sulphur</p>