

Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Mark the correct statement.	A. Na ⁺ is smaller than Na atom B. Na ⁺ is large than Na atom C. Cl ⁻ is smaller than Cl atom D. Cl ⁻ and Cl are equal in size
2	Which element when react with chlorine form polymeric halide.	A. Na B. Be C. Ba D. P
3	Variable valency is generally exhibited by	A. Transition elements B. Alkali metals C. s-block elements D. Gaseous elements
4	Which has greater hydration energy.	A. Li ⁺ B. Na ⁺ C. K ⁺ D. Mg ⁺²
5	Which of the following statement is correct	A. Na atom is smaller than Na ⁺ B. Na atom is larger than K atom C. F atom is smaller than F ⁻ D. F atom is larger than F ⁻
6	Keeping in view size of atoms ,which order is the correct one.	A. Mg > Sr B. Ba > Mg C. Lu > Ce D. Cl > I
7	In which group, melting point and boiling point increase downward in a group	A. IA B. II A C. VII A D. Both a and b
8	Which class of elements shows low value of first ionization potential.	A. Alkali metals B. Alkaline earth metals C. Halogens D. Noble gases
9	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
10	Transition elements are those	A. Which comes before uranium B. Which comes after uranium C. Which are prepared artificially D. Which are in between s-block and p - block elements
11	In modern periodic table VI period contains elements	A. 8 B. 18 C. 10 D. 32
12	Across a period from left to right in the periodic table, the melting and boiling point.	A. Decrease B. Increase C. Remain constant D. First increase upto the middle of period and then decrease
13	Which one of the following sets has coinage metals is it.	A. Cu, Hg, Au B. Cu, Ag, Au C. Ag, Au, Hg D. Cu, Fe, Au
14	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F

15	Mark the incorrect statement	A. Metallic character increase down the group B. Metallic character increase from left to right along a period C. Metallic character decrease from left to right along a period D. Metallic character remains the same down the group
16	Which one of the following elements burns in air to form an oxide which, when shaken with water, give a solution with a pH greater than 7.	A. Carbon B. Magnesium C. sulphur D. Hydrogen
17	Fluorine is in group VII A of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. Iodine D. Sulphur
18	What are the total numbers of periods in the modern periodic table.	A. 3 B. 5 C. 7 D. 8
19	Which one of the following is a metalloid.	A. sulphur B. Antimony C. Mercury D. Zinc
20	Which of the following element has lowest ionization energy	A. Beryllium B. Boron C. Carbon D. Oxygen