Periodic Table and Periodic Properties

Sr	Questions	Answers Choice
1	Peroxides are those in whcih oxidation state of oxygen is	A. -1 B. +2 D. -2 E. -1/2
2	Metallic oxides are.....in nature.	A. Acidic B. Basic C. Amphotric D. Depends upon oxide
3	Which oxidation state will not shown by S	A. -2 B. +4 C. +8 D. +2
4	L- Pauling developed a scale for	A. Ionization energy B. Electron affinity C. Electronegativity D. None
5	Which set of the followig conditions results in higher ionization energy.	A. Smaller atom and greater nuclear charge B. Smaller atom and smaller nuclear charge C. Larger atom and greater nuclear charge D. Large atom and the smaller nuclear charge
6	Acidic oxides are	A. Metallic oxides B. Non Metallic oxides C. Metalloid oxides D. All
7	Which property increases as you go down a group in the periodic table?	A. Atomic radium B. Elecron affinity C. Electronegativity D. Ionization Energy
8	Electronegativity value of "Cs" is	A. 2.1 B. 0.79 C. 0.1 D. 4
9	Who considere as the father of the periodic table	A. Moseley B. Mendeleev C. Newland D. Dobereiner
10	Which element is used in S.O.S flares and in firework.	A. Mg Powder B. Na powder C. CaCO3 D. KO2
11	John Newlands gavt he idea of.	A. Law of triads B. Law of Octaves C. Modern Periodic law D. Curves between weight and volume
12	Which one do you think is correct about metallic character.	A. It decrease frin tio ti vittin ub a griyo B. It increases from top to bottom in a group C. It remians constant fro left to right in a period D. It increases from left to right in a period
13	Element that can show variable oxidation states.	A. Mg B. Ca C. S D. Mg
		A. More than 7

14	Acidic chlorides have pH?	B. Less than 7 C. Equal to 7 D. May be any
15	Least electronetative element is.	A. Cs B. H C. Li D. He
16	Select metalloid from the following.	A. Po B. Cl C. Fe D. P
17	What is teh oxidation state of sulfur in the sulfate ion $SO^{-2}4$	A. +4 B. +6 C. +2 D. 0
18	Which one of the following element has the highest ionization energy.	A. Argon B. Sodium C. Aluminium D. Magnesium
19	Na burns with oxygen to give.	A. Golden yellow flame B. Golden brown flame C. Crimson red flame D. Intense white flame
20	Which of the following has highest ionization energy.	A. P B. S C. Ar D. Cl