

Periodic Table and Periodicity

Sr	Questions	Answers Choice
1	The electronegativity of carbon is	A. 2.5 B. 2.0 C. 1.0 D. 4.0
2	Which one of the followign halogns has highest electronegativity	A. Iodine B. Fluorine C. Chlorine D. Bromine
3	Metals can form ions carrying carges.	A. Uni positive B. Di positive C. Tri positive D. All of these
4	4th and 5th priod of the long form of periodic table are called.	A. Short periods B. Normal periods C. Very long peiods D. Long periods
5	Metals posses.	A. Ionic bond B. Covalent bond C. Co-ordinate covalent D. Metallic bond
6	How many elements are prsent in each 2nd and 3rd period.	A. 2 B. 32 C. 18 D. 8
7	How many blocks are presnt in modern periodic table	A. 2 B. 3 C. 4 D. 5
8	Whcih period of the modern periodic table is considered as incomplete period.	A. 4th B. 5th C. 6th D. 7th
9	Actinidws belong to period.	A. 4th B. 5th C. 6th D. 7th
10	/which of the followign elements is presnt in group IA.	A. Lithium B. Hydrogen C. Sodium D. All of these
11	How many electrons are present in the valence shell of group 1 elements.	A. 1 B. 2 C. 3 D. 4
12	The amouth of energy given out when an electron is added to an eatom is called.	A. Electron affinity B. Lattice energy C. Ionization energy D. Electronegativity
13	Which group elements has low ionization energies.	A. Halogens B. Noble gases C. Alkaline Earth Metals D. Alkali Metals
14	The elementss of group 1 and 2 are placed in which block	A. s B. p C. d D. f
15	Actinide series starts after the element	A. Actinium B. Lanthanum C. Osmum D. Silver

16	Along the period which one of the following decreases.	A. Electronegativity B. Ionization energy C. Atomic radius D. Electron affinity
17	The ability of an atom to attract the shared pair of electrons towards itself in a molecule is called	A. Ionization energy B. Electronegativity C. Shielding effect D. Electron affinity
18	The elements of group 3 to 12 are called.	A. Normal elements B. Halogens C. Noble gases D. Transition elements
19	Which halogen element exists in a liquid state at room temperature	A. Bromine B. Chlorine C. Fluorine D. Iodine
20	A yellow solid element exists in allotropic forms which is also present in fossil fuel. Indicate the name	A. Iodine B. Carbon C. Sulphur D. Aluminium