

## Periodic Table and Periodicity

Sr	Questions	Answers Choice
1	The haviest metal is	A. Iron B. Lead C. Osmium D. Platinum
2	Elements o group 13 to 18 have thier valence electrons is subshell	A. s B. p C. f D. d
3	Which is strongest oxidizing agent.	A. Cholorine B. Iodine C. Fluorine D. Bromine
4	The electronegativity of carbon is	A. 2.5 B. 2.0 C. 1.0 D. 4.0
5	Atomic number of actinium is	A. 57 B. 60 C. 89 D. 80
6	How many elements are present in 7th period.	A. 2 B. 8 C. 18 D. 23
7	The element having less value of ionizatin energy and less value of electron affinity is likely to belong to.	A. Group1 B. Group 13 C. Group 16 D. Group 17
8	Electronegativity of oxygen is.	A. 3,1 B. 3,3 C. 3.4 D. 3,2
9	Which of the following grops contain alkaline earth metals.	A. I A B. II A C. VII A D. VIII A
10	4th and 5th priod of the long form of periodic table are called.	A. Short periods B. Normal periods C. Very long peiods D. Long periods
11	Actinidws belong to period.	A. 4th B. 5th C. 6th D. 7th
12	How many elements are present in 6th period.	A. 2 B. 8 C. 18 D. 32
13	Actinide series starts after the element	A. Actinium B. Lanthanum C. Osmum D. Silver
14	Metals posses.	A. Ionic bond B. Covalent bond C. Co-ordinate covalent D. Metallic bond
15	How many blocks are presnt in modern periodic table	A. 2 B. 3 C. 4 D. 5

16	Which one of the following halogens has lowest electronegativity	A. Iodine B. Chlorine C. Fluorine D. Bromine
17	Which element has the highest melting point.	A. K B. Cs C. Na D. Rb
18	Which is the softest metal.	A. Zn B. Ca C. Na D. Al
19	How many periods are present in the modern periodic table	A. 7 B. 8 C. 10 D. 12
20	When we move from left to right in a period, ionization energy.	A. No effect B. Decreases C. Increases D. None of these